

Conceptus.

Corrosion Susceptibility on Conceptus Essure Micro-insert: 6 Month Report

VR-0191.PV.Rev2

Approvals:

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Total of 85 pages
PW 2-17-04

1.0 SUMMARY

- The Essure Micro-insert passes the corrosion susceptibility bench test.
- 54 Essure Micro-inserts were tested for up to six months in a worst-case bench environment.
- The daily leaching rate of nickel and tin ions released are at least 2000 times less than everyday intake of food and water and exposure to environment.
- The daily leaching rate of chromium is below the detection limit.
- The Essure Micro-insert maintained mechanical integrity during the six months of exposure to a corrosive saline environment.

2.0 PURPOSE

The purpose of the corrosion bench test was to measure the levels of potentially harmful metal ions released, and assess loss of mechanical integrity and overall corrosion damage to Essure Micro-inserts over a six-month period in a worst case, corrosive bench-top environment.

3.0 SCOPE

- 3.1 This report covers all six months of the corrosion bench test. The first revision of this report was an interim report covering only the first three months of data.
- 3.2 The test articles were Essure Micro-inserts, internally known as Device Subassemblies, E0648-02.

4.0 REFERENCE DOCUMENTS

- | | | |
|-----|-------------------|---|
| 4.1 | E0648-02 | Device Subassembly |
| 4.2 | VP-0191.PV, Rev 1 | Protocol corresponding to this validation |
| 4.3 | VP-0191.PV, A01 | Addendum to protocol |

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4.4	VR-0191.PV, Rev 1	3 month interim report corresponding to this validation
4.5	VR-0213.PV, Rev2	Corrosion Properties of the Essure Implant: X-Ray Examination Evaluation
4.6	SOP-01046, Rev K	Validation Policy
4.7	DR 069, Rev 0	Verification Bench Testing
4.8	DR 047, Rev 0	Summary Report of Essure Micro-insert Corrosion Evaluation
4.9	VP-0246.PV.Rev1	Product Validation Protocol: Biocompatibility Testing of the Essure System
4.10	VR-0246.PV.Rev1	Product Validation Report: Biocompatibility Testing of the Essure System

5.0 ATTACHMENTS AND APPENDICES.

5.1	Appendix 1	Table of Cumulative Leaching Rates
5.2	Appendix 2	Table of Disruptive Leaching Rates
5.3	Appendix 3	Build records for test samples
5.4	Appendix 4	Documentation of procedure (copies of lab notebook pages)
5.5	Appendix 5	Leaching analysis data sheets.
5.6	Attachment 1	VP-0191.PV.Rev1 (copy)
5.7	Attachment 2	VP-0191.PV A01 (copy)

6.0 BACKGROUND

No standard exists for corrosion testing of medical devices intended for implantation in the fallopian tubes. Therefore, this study was designed for the specific requirements of the Essure Micro-insert. The results of this study should be considered together with the results of clinical evaluations to determine the corrosion susceptibility of the device. See DR047, Rev 0.

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6.1 Leaching Rate of Metal Ions

This bench test is designed to measure the leaching rate of potentially harmful metal ions. The Essure Micro-insert is made of the materials listed in Table 1 below.

Table 1: Material composition of components of the Essure Micro-insert

Components	Material	Material Composition
Outer Coil	Chromium doped Nitinol (Nickel-Titanium)	56% Nickel, 44% Titanium, trace Chromium
Inner Coil, Thread Coil	316L stainless steel	apx. 62.5% Iron, 17.6% Chromium, 14.5% Nickel
Stopper Band, Platinum Band, Platinum Ring, Bump	Platinum-Iridium	90% Platinum, 10% Iridium
Solder	Tin-Silver	95% Tin, 5% Silver
Fiber	PET polyester, white	≥ 92% Polyester, < 5% TiO ₂ , < 3% fiber lubricants

From this list of materials, three metal ions have been identified as potentially harmful: nickel, chromium, and tin. Nickel is of most concern, because exposure to nickel is known to cause nickel allergy. The leaching rates of other metals were not measured, because they are not known to have harmful effects on the human body.

This study seeks to quantify the leaching rate of the three metal ions of concern and compare them to known human intake levels. The study is not designed to determine the effects of corrosion products on the human body. Instead, the biological effects of implanting the Essure Micro-insert have been addressed through biocompatibility testing as described in VP-0246.PV.Rev1 and VR-0246.PV.Rev1.

6.2 Mechanical Integrity of Micro-insert

This corrosion study is also designed to determine whether or not the mechanical integrity of the Micro-inserts is maintained after exposure to a

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corrosive environment for three months. During the first three months of implantation, the outer coil holds the inner coil and polyester fibers in place while tissue grows into the Micro-insert. After tissue ingrowth has occurred, the Micro-insert is held in place by tissue. If corrosion should cause the inner coil and outer coil to completely come apart during the first three months of implantation, and then a subsequent migration of one of the detached components were to occur, it is possible that the Micro-insert may not be held in place long enough to stimulate complete tissue ingrowth and blockage of the fallopian tube. The Micro-insert could not be relied on for contraception if this situation occurred.

6.3 General Corrosion Susceptibility

In addition, the SEM images created as part of this study provide highly detailed images that can be examined for signs of unexpected corrosion.

6.4 Bench Top Corrosion Environment

The bench corrosion environment was designed to be potentially more corrosive than the *in-vivo* environment. The implants were maintained in 37° C physiological saline similar to the environment of the fallopian tubes. The saline was buffered to be slightly acidic, because the uterus near the fallopian tubes is known to have a pH from 6.3 to 7.1^{1,2}. It was not known whether corrosion would be worse in an enclosed environment or an environment flushed by saline, so samples were tested in two different ways. One group of samples was left in saline for up to 180 days. Another group of samples was also tested for 180 days but had the saline periodically replaced with fresh saline.

¹ Feo L.G. (1955). "The pH of the Human Uterine Cavity *In Situ*"; *Am. J. Obst. & Gynec.* 60-64

² Kudla T. (1964). "The Evaluation of the Hydrogen Potentials (pH) of the Uterine Cavity *In Situ* in the Course of the Menstrual Cycle in Comparison with Results of Endometrial Biopsies and the Measurements of the Basal Body Temperature."

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The X-ray evaluation of clinical data from the Phase II and Pivotal trials showed no measurable loss of solder due to corrosion as compared to visible loss of solder in the bench study, validating the design of the bench study to simulate a potentially more corrosive environment on the bench *in-vivo*. See the three month summary report, DR047, Rev 0.

7.0 PROCEDURE

7.1 Samples

7.1.1 Samples were Device Subassemblies E0648-02, lot # RD0127001. Build documentation is included in Appendix 3.

7.1.2 54 Device Subassemblies were tested. Two Device Subassemblies were tested per vial to model the number of implants one patient would receive.

7.1.3 6 Device Subassemblies (in three vials) were used as controls.

7.2 Procedure

The corrosion bench study was performed according to the procedure in VP-0191.PV.Rev1 and A01. Refer to Attachments 1 and 2.

Documentation of the procedure was recorded in lab notebooks.

Photocopies of the lab notebook pages are included in Appendix 4.

Protocol deviations are explained in Section 10.

Two types of tests were performed on the samples. Group A "cumulative" samples were placed in saline in a heated water bath for periods of time ranging from 7 days to 180 days. At the specified time, the vials were removed from the heated water bath, the saline was sent out for analysis, and the Micro-inserts were cleaned and examined. Group B "disruptive" samples were placed in saline in a heated water bath. At specified times, the saline was removed for analysis, and fresh saline was added to the vials. At 180 days, the vials were removed from the heated water bath, the remaining saline was sent out for analysis, and the Micro-inserts were



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cleaned and examined. Figure 1 shows the time each sample remained in saline.

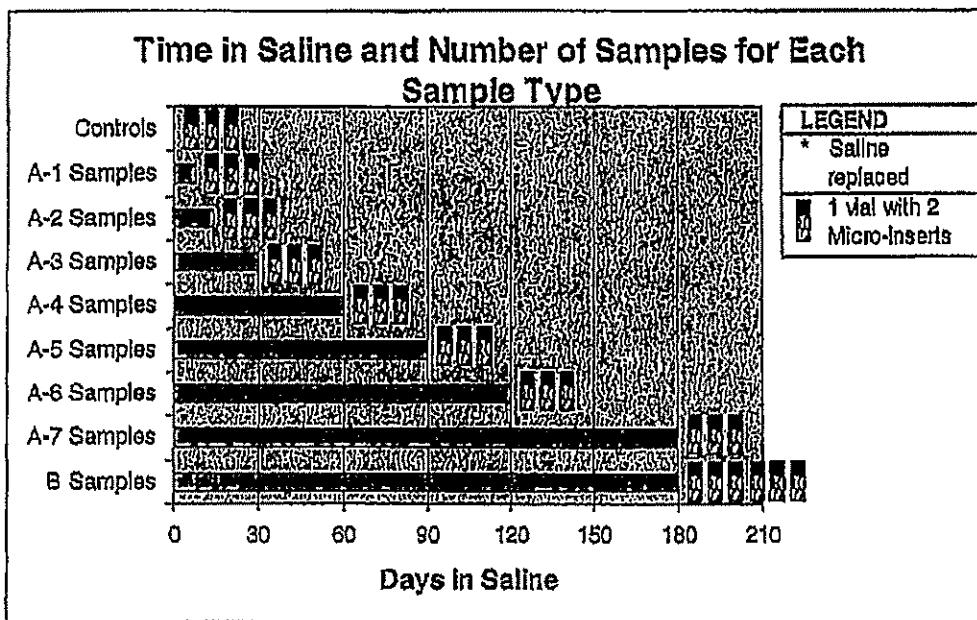


Figure 1: Graphical description of number of samples and days in saline.

7.3 Equipment

Only calibrated equipment was used, except in the case of the heated water bath. Because the water bath was not calibrated, a calibrated electronic temperature gage was used to monitor the water bath temperature. See the protocol addendum located in Attachment 2.

- 7.3.1 Heated water bath
- 7.3.2 Electronic temperature gage
- 7.3.3 Ultrasonic cleaner
- 7.3.4 Microscope
- 7.3.5 pH test strips

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7.4 Personnel

- 7.4.1 Sample preparation and handling was performed by R&D engineers, Christian Lowe and Elisa Aldridge, and by R&D technician, Cheryl Swanson.
- 7.4.2 SEM imaging was performed by SEM scientist Jane Wheeler at Charles Evans & Associates.
- 7.4.3 Metal leaching analysis was performed by analyst DH at Scientific Environmental Laboratories.
- 7.4.4 Jay Yang, PhD of Global Nitinol Technologies was consulted to assist in designing the study and interpreting the results. Jay Yang wrote the protocol and three month report, VP-0191.PV.Rev1 and VR-0191.PV.Rev1.

8.0 ACCEPTANCE CRITERIA

8.1 Leaching Rate of Nickel

The leaching rate of nickel ions from the samples must be lower than the average levels of human intake of nickel from diet and the environment.

8.2 Mechanical Integrity

The Micro-inserts must maintain mechanical integrity for at least three months. That is, each Micro-insert must still be in one piece after exposure to a corrosive saline environment for three months. In particular, the fibered inner coil must remain attached to the outer coil.

9.0 RESULTS AND DISCUSSION

The test passed both acceptance criteria. The leaching rate of nickel and tin ions released due to corrosion were at least 2000 times lower than the daily human intake of these metal ions from the diet and environment. The leaching rate of chromium was below the detection limit. All of the samples tested maintained mechanical integrity, not just for three months, but for all six months of the study.



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9.1 Metal Ion Leaching Analysis Summary

The levels of nickel and tin ions released into the solution were found to be at least 2000 times below the average levels of human intake of these ions from diet and the environment³. The level of chromium released was below the detection limit. According to published data, the normal human intake of nickel and tin from food and water and the observed highest measured leaching rates for the three metals monitored are as follows:

Table 2: Highest Metal Leaching Rates Compared to Normal Human Intake

Metal	Normal Human Daily Intake ³	Highest Measured Leaching Rates
Nickel	300 µg/day	0.14 µg/day
Tin	100,000 µg/day	27 µg/day
Chromium	no published data is available	less than 0.03 µg/day (below detection limit)

9.1.1 Metal Ion Leaching Analysis Cumulative Data

Figure 2 and Figure 3 show the leaching rate of nickel and tin as analyzed in the cumulative test (A samples). The leaching rate of chromium was below the detection limit at all time points. The test was an evaluation of the cumulative metal ion release at each time point (7 days, 14 days, 30 days, 60 days, 90 days, 120 days, and 180 days). The samples were placed in a vial of saline and left in a heated water bath for the number of days indicated, and the saline was removed and analyzed at the end of the time period indicated. Table 3 containing the calculated data is located in Appendix 1. The raw data is located in Appendix 5.

³ Medical and Biological Effects of Environmental Pollutants, published by the National Academy of Sciences, Washington, D.C., 1975.

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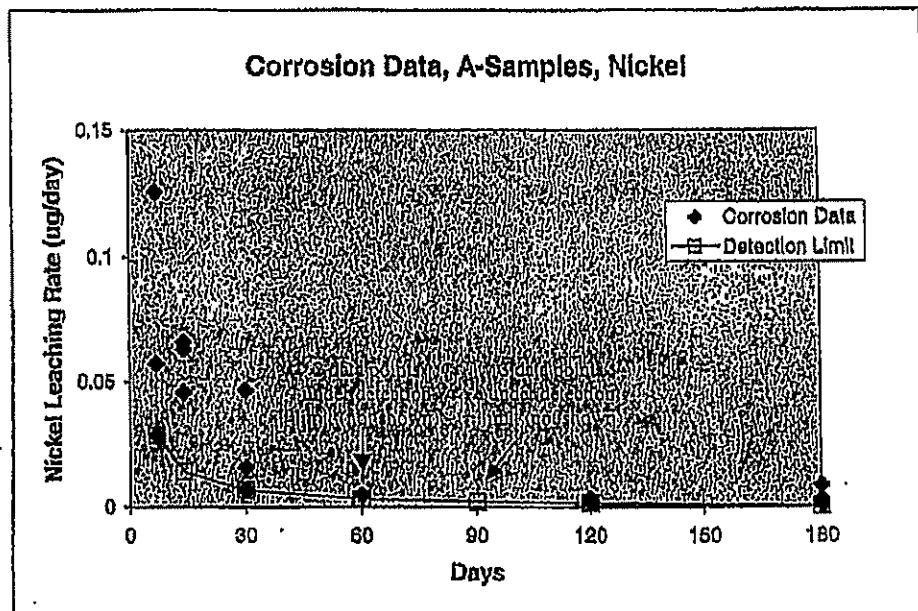


Figure 2: Leaching rate of Nickel for Cumulative Samples.

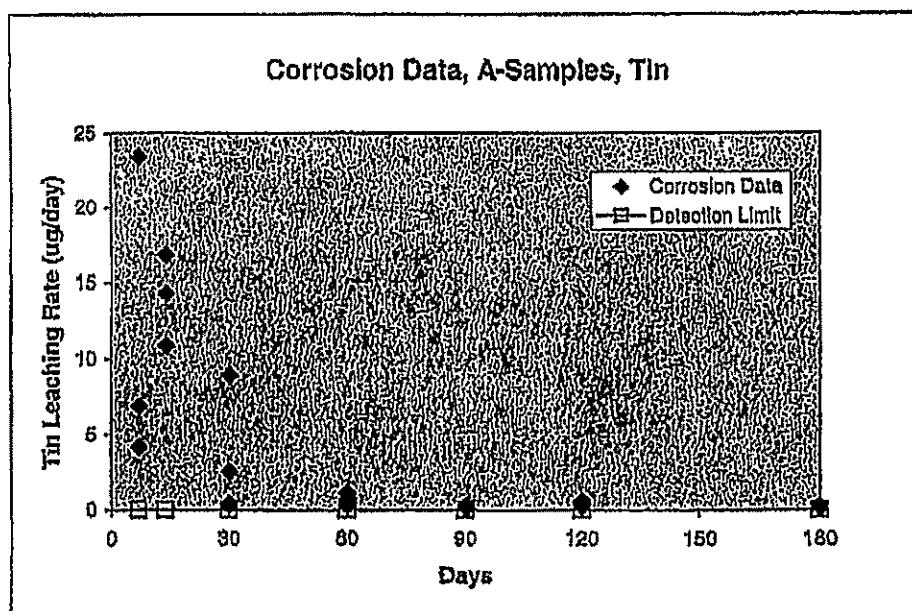


Figure 3: Leaching rate of Tin for Cumulative Samples.



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9.1.2 Metal Ion Leaching Analysis Disruptive Data

Figure 4 and Figure 5 show the leaching rate of nickel and tin as analyzed in the disruptive test (B samples). The leaching rate of nickel was below the detection limit at all time points except 7 days and 180 days. The leaching rate of chromium was below the detection limit at all time points. The test evaluated metal ion release for 0-7 days, 7-14 days, 14-30 days, 30-60 days, 60-90 days, 90-120 days, and 120-180 days, for a single set of samples and was termed the "disruptive test". At each time indicated, the saline was removed for analysis, and fresh saline was added to the vials containing the Micro-insert samples. The data for the "disruptive test" is included in Table 4 in Appendix 2. Raw data is located in Appendix 5.

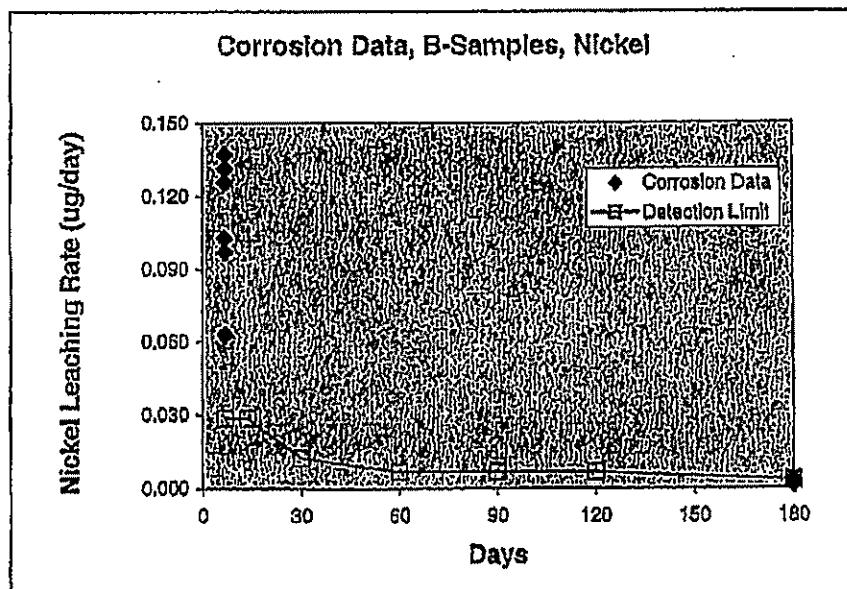


Figure 4: Leaching rate of Nickel for Disruptive Samples.

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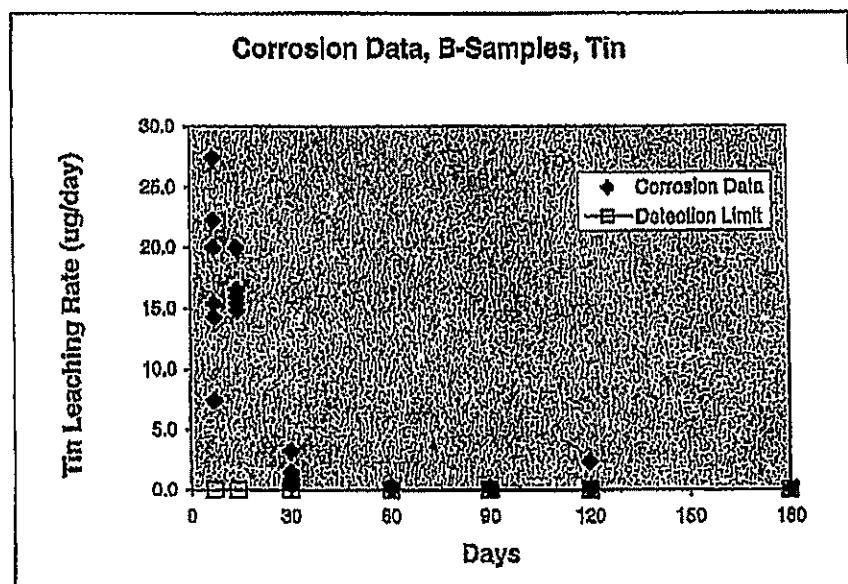


Figure 5: Leaching rate of Tin for Disruptive Samples.

9.2 Mechanical Integrity

The samples passed the acceptance criteria for mechanical integrity. All of the samples tested maintained mechanical integrity, not just for three months, but for all six months of the study.

After the Micro-insert samples were removed from the vials of solution, they were cleaned in an ultrasonic bath, dried, and examined using scanning electron microscopy (SEM). No Micro-inserts showed loss of mechanical integrity. In all cases, the outer coil remained attached to the fibered inner coil.

9.3 Examination for Signs of Corrosion

As expected, the solder showed signs of corrosion resulting in pitting and increasing porosity with the worst corrosion damage on the ball tip. At the three-month time point, approximately 25-50% of the solder had corroded.

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At the six-month time point, the ball tips of some of the samples were almost completely corroded, but all of the solder bonds continued to hold together. In all cases, the outer coil remained attached to the fibered inner coil. This is an acceptable level of solder corrosion, because it did not result in the loss of mechanical integrity. No other components showed signs of corrosion.

10.0 DEVIATIONS

- 10.1 In the Scope section of the protocol, one goal of the study was described as "determine minimum time needed for structural disintegration /detachment in a simulated body environment." This goal was not achieved, since none of the samples showed loss of mechanical integrity during the 180 days of the study. The fact that none of the samples lost mechanical integrity is considered a favorable outcome, and a new study will not be conducted to learn the time needed for loss of mechanical integrity.
- 10.2 The protocol specified that the temperature of the water bath be checked at least once per week. However, the week of December 24, 2001, the water bath temperature was not checked. This minor deviation did not affect the results of the study.
- 10.3 The protocol says to ultrasonically clean the samples before and after SEM handling. None of the devices was cleaned after SEM imaging. This deviation had no affect on the study results.
- 10.4 The protocol requires two reports to be written: one at two months, and a final report with all six months of data. A three month report was written, rather than a two month report, because three months of data were submitted to the FDA in the PMA Module III: Biocompatibility. This deviation has no affect on the study results.
- 10.5 The pH of the six month saline samples was checked using pH test strips before the saline samples were sent out for metal leaching analysis.

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Testing the pH was not part of the original protocol. This deviation does not affect the study results.

11.0 CONCLUSIONS

- The Essure Micro-insert passes the corrosion susceptibility bench test.
- The daily leaching rate of nickel and tin ions released are at least 2000 times less than everyday intake of food and water and exposure to environment.
- The daily leaching rate of chromium is below the detection limit.
- The Essure Micro-insert maintained mechanical integrity during the six months of exposure to a bench top corrosive saline environment.
- The corrosion rate of the solder is acceptably low.
- Besides the solder, no other signs of corrosion were visible in the SEM images.

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APPENDIX I

APPENDIX 1

Table 3: Cumulative Simulated Environment (A Samples)

Sample #	Time in Saline (days)	Nickel ($\mu\text{g}/\text{day}$)	Chromium ($\mu\text{g}/\text{day}$)	Tin ($\mu\text{g}/\text{day}$)
A-1-a	7	0.03	<0.03	4.2
A-1-b	7	0.13	<0.03	23
A-1-c	7	0.057	<0.03	6.9
<i>Average per vial</i>		0.07	0.03	11
A-2-a	14	0.066	<0.01	11
A-2-b	14	0.046	<0.01	17
A-2-c	14	0.063	<0.01	14
<i>Average per vial</i>		0.058	0.01	14
A-3-a	30	0.007	<0.007	0.37
A-3-b	30	0.047	<0.007	8.9
A-3-c	30	0.016	<0.007	2.5
<i>Average per vial</i>		0.023	0.007	3.9
A-4-a	60	0.005	<0.003	1.09
A-4-b	60	<0.003	<0.003	0.720
A-4-c	60	<0.003	<0.003	0.28
<i>Average per vial</i>		0.004	0.003	0.70
A-5-a	90	<0.002	<0.002	0.098
A-5-b	90	<0.002	<0.002	0.20
A-5-c	90	<0.002	<0.002	0.20
<i>Average per vial</i>		0.002	0.002	0.17
A-6-a	120	0.0033	<0.002	0.587
A-6-b	120	0.002	<0.002	0.23
A-6-c	120	0.002	<0.002	0.21
<i>Average per vial</i>		0.002	0.002	0.34
A-7-a	180	0.0009	<0.001	0.12
A-7-b	180	0.0031	<0.001	0.13
A-7-c	180	0.0089	<0.001	0.258
<i>Average per vial</i>		0.0043	0.001	0.17

The raw data is measured in (mg/L). To calculate the rate in ($\mu\text{g}/\text{day}$) the following equation was used:

$$\text{rate}(\mu\text{g}/\text{L}) = \frac{\text{rate}(\text{mg}/\text{L}) * (4\text{mL}) * (1000\mu\text{g}/\text{mg})}{(1000\text{mL}/\text{L}) * (\text{days_in_saline})}$$

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The number of significant digits shown is the maximum number of significant digits possible based on the raw data.

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APPENDIX 2

APPENDIX 2

Table 4: Disrupted Simulated Environment (B Samples)⁴

Sample #	Time Device in Saline (days)	Time Saline Tested (days)	Nickel (µg/day)	Chromium (µg/day)	Tin (µg/day)
B-1-a	7	7	0.097	<0.03	14
B-1-b	7	7	0.14	<0.03	20
B-1-c	7	7	0.13	<0.03	22
B-1-d	7	7	0.13	<0.03	27
B-1-e	7	7	0.10	<0.03	15
B-1-f	7	7	0.063	<0.03	7.4
<i>Average per vial</i>			0.11	0.03	18
B-2-a	14	7	<0.03	<0.03	15
B-2-b	14	7	<0.03	<0.03	16
B-2-c	14	7	<0.03	<0.03	15
B-2-d	14	7	<0.03	<0.03	20
B-2-e	14	7	<0.03	<0.03	17
B-2-f	14	7	<0.03	<0.03	20
<i>Average per vial</i>			0.03	0.03	17
B-3-a	30	16	<0.01	<0.01	3.3
B-3-b	30	16	<0.01	<0.01	0.50
B-3-c	30	16	<0.01	<0.01	0.90
B-3-d	30	16	<0.01	<0.01	0.45
B-3-e	30	16	<0.01	<0.01	0.88
B-3-f	30	16	<0.01	<0.01	1.5
<i>Average per vial</i>			0.01	0.01	1.3
B-4-a	60	30	<0.007	<0.007	0.40
B-4-b	60	30	<0.007	<0.007	0.16
B-4-c	60	30	<0.007	<0.007	0.23
B-4-d	60	30	<0.007	<0.007	0.43
B-4-e	60	30	<0.007	<0.007	0.64
B-4-f	60	30	<0.007	<0.007	0.36
<i>Average per vial</i>			0.007	0.007	0.37

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⁴ Some of the data for the disruptive samples that were reported to the FDA in Table 2 of Appendix F of the June 21, 2002 response were in error. However, the true values are even lower than those that were erroneously reported. This table includes the corrected values.

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Sample #	Time Device in Saline (days)	Time Saline Tested (days)	Nickel (µg/day)	Chromium (µg/day)	Tin (µg/day)
B-5-a	90	30	<0.007	<0.007	0.68
B-5-b	90	30	<0.007	<0.007	0.19
B-5-c	90	30	<0.007	<0.007	0.07
B-5-d	90	30	<0.007	<0.007	0.09
B-5-e	90	30	<0.007	<0.007	0.1
B-5-f	90	30	<0.007	<0.007	0.27
<i>Average per vial</i>			0.007	0.007	0.23
B-6-a	120	30	<0.007	<0.007	0.092
B-6-b	120	30	<0.007	<0.007	0.044
B-6-c	120	30	<0.007	<0.007	<0.007
B-6-d	120	30	<0.007	<0.007	0.020
B-6-e	120	30	<0.007	<0.007	0.081
B-6-f	120	30	<0.007	<0.007	2.35
<i>Average per vial</i>			0.007	0.007	0.43
B-7-a	180	60	0.002	<0.003	0.006
B-7-b	180	60	0.003	<0.003	0.003
B-7-c	180	60	0.001	<0.003	0.005
B-7-d	180	60	0.003	<0.003	0.003
B-7-e	180	60	0.003	<0.003	0.0093
B-7-f	180	60	0.002	<0.003	0.040
<i>Average per vial</i>			0.002	0.003	0.011

The raw data is measured in (mg/L). To calculate the rate in (µg/day) the following equation was used:

$$\text{rate}(\mu\text{g/L}) = \frac{\text{rate}(\text{mg/L}) * (4\text{ml}) * (1000\mu\text{g/mg})}{(1000\text{ml/L}) * (\text{days_saline_tested})}$$

The number of significant digits shown is the maximum number of significant digits possible based on the raw data.

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APPENDIX 3

CONTINUATION CORROSION STUDY

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TRAVELER for
STOP, Sterile, 2-Device - 100959 (E1635)

Shop Order 1058
Traveler no. 1/1
BPCS entry

Process	Materials, BPCS #		Initial	Date	Qty Passed
1. Equipment Setup	N/A	N/A	EC	10/10/01	N/A
2. Line Clearance SOP00922 rev (E)	N/A	N/A	EC	10/10/01	N/A
3. QA Verification of Line Clearance SOP00922 rev (E)	N/A	N/A	HS	10/10/01	N/A
4. Fiber R0687 rev (C)	Fiber 100913 Inner/Outer Coll Subassembly 101029 70% Alcohol 100177	Lot no. <u>1118000150 / 0103000005</u> Lot no. <u>D-2127709</u> Lot no. <u>0105000057 Exp 11/31/01</u>	TK TTV	10/10/01 10/10/01	66
5. Wind Down R1695 rev () E0685-03 rev ()	Release Cath S/A 100828 Delivery Wire S/A 100798 OW Holder 100730 Set Screw 100793	Lot no. Lot no. Lot no. Lot no.			SEND NOTE
6. Inspection I Q0685-03 rev () VS1590 rev ()	Inspection Notes:	Qty reworked: _____ by _____ on / /			
7. Bend Tip R1502 rev ()	N/A	N/A			
8. Handle I R1502 rev ()	Delivery Cath S/A 100886 Loclite 4011 100498 Right Handle 100728 Rack 100728 Otent 100725 Thumbwheel 100724 Button 100723	Lot no. Lot no. Lot no. exp. Lot no. Lot no. Lot no. Lot no.			
9. Inspection I QD788 rev () E0788 rev () VS1534 rev ()	Inspection Notes:	Qty reworked: _____ by _____ on / /			
10. Handle II R1502 rev ()	Left Handle 100727 S. S. Screws 100800	Lot no. Lot no.			
11. Inspection II Q0766 rev ()	Inspection Notes:				
12. Package R1617 rev ()	Single Tray S/A 101005 Pouch 100822 Sticker 100128	Lot no. Lot no. Lot no.			
13. Seal Pouches R0102 rev ()	N/A	N/A			
14. Inspection II Q1624 rev ()	Inspection Notes:	Qty reworked: _____ by _____ on / /			

Additional Traveler Notes: FOUR ASSEMBLIES (101029) FROM
SHOP ORDER 1107 WERE RESTRICTED BUT HAD THE FIBER
PROCESS PERFORMED AS PART OF THIS SHOP ORDER FOR
A TOTAL OF 70 DEVICES.

CONTROLLED DOCUMENT

10/10/01

Revision Level
J

Conceptus CONFIDENTIAL Document
QAF-1700 LHR for 100959 (E1635) - Traveler for STOP, Sterile 2-Device

Page 1 of 1

V R 0791

Lot History Record Review Form
QAF-00925

Part Number E0644-02 rev. C
 Lot Number RD 0127710
 Shop Order No. 1108-1

General

- Input fields are completed with correct information (signatures, initials, dates, quantities, data).
- All pages are present.

PASS FAIL

Inspection and Testing

- Verify that inspections were performed at the correct sample size.
- Verify samples have passed any required testing.

PASS FAIL

Revision Levels

- Revision levels of all documents used are recorded.
- Verify that revisions recorded are appropriate for build dates.

PASS FAIL

Raw Material Lot Numbers

- All Raw Material lots used have been recorded.
- Raw Material lots are released prior to LHR closure.
- Raw Material lots have not expired.

PASS FAIL

Quantity

- Are changes in quantity reasonable?
- Quantities cannot increase without justification.

PASS FAIL

Comments

- Comments must be signed and dated.
- Review all comments recorded on the front and back of the LHR.
- Verify that comments make sense and resulting actions are appropriate.
- If comments refer to material being rejected, were units rejected?

PASS FAIL

Addition requirements for Finished Goods or Top Assemblies

- Verify that Subassembly lots are released prior to LHR closure.
- Sterilization Date YEAR MONTH DAY (List the date Aeration is complete). Sterile Load SL-
- Expiration Date YEAR MONTH DAY (Required for Sterile Labeled Finished Goods)

PASS FAIL

Step	Discrepancy	Resolved by	Date	Comments
8-15	<u>Line clearance: Not performed</u>	<u>CL/KY</u>	<u>12/10/01</u>	<u>R&D accepts lot as is? see NM note to CL (12/10/01) regarding purpose of lot</u>

If applicable, the following must be completed:

Red-lines to an ECO-controlled process require a completed ECO prior to LHR release.
 Attach the ECO cover page for reference.

List document numbers and new rev level.

MRRs or QAs must be dispositioned prior to LHR release. Any unusual observations noted were investigated via MRR or QA. Attach verification.

List MRR or QA numbers

801

QUANTITY DISPOSITIONED <u>70</u>		REVIEWED BY <u>[Signature]</u>	DATE <u>12/10/01</u>
Disposition Assigned:		Operations	Doc Control
<input checked="" type="checkbox"/> Approved <input type="checkbox"/> Rejected <input type="checkbox"/> Other _____		<input type="checkbox"/> D-151	
If Rejected, designate one or more follow-up actions:		DISPOSITIONED BY <u>Kirk J. Young</u> DATE <u>12/10/01</u> Quality Engineer or Designee	
<input type="checkbox"/> Rework <input type="checkbox"/> Stop <input type="checkbox"/> Non-Human Use Only		DISPOSITIONED BY <u>Kirk J. Young</u> DATE <u>12/10/01</u> Quality Engineer or Designee	

QAF-00925 Rev. H (Refer to SOP-00921)

VR 0191

MEMO

Date: 12/12/01
To: C. Lowe
From: M. Martinez *M. Martinez 12/12/01*
Re: Etched units used for corrosion test.

The following describes the process and inspections performed on etched outer coils used for VP-0191.PV Rev 1

1. Three lots of etched outer coils were produced. These Lots were entitled Lots A, B and C. The LHR's used to produce Etched Coils from Lots A,B, and C are attached to this memo. Only steps 4 through 6 were performed per the QAF-1701 rev E, Outer coil subassembly. All raw materials are recorded on these LHR's.
2. The etching process was performed on all three lots following the attached redline Oxide Removal work instruction R0704, and attached redline Work instruction Acid Preparation R1727.
3. Inspection of coils were performed per attached redlined, inspection was performed by QA. The lots were identified by the Test group A and B, followed by Lot group A,B, or C.
4. 80 coils were handed to Manufacturing for use in engineering build for Corrosion testing. These 80 coils were randomly pulled from all three of the lots A, B, C. No other process or manipulation was performed on the coils before the transfer.



VR 0191

Attribute Data Sheet

Sample	Pass	Fail	Comments
1		X	
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X		
19	X		
20	X		
21	X		
22	X		
23	X		
24	X		
25	X		
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39			
40			

Date: 9-21-01

Performed By: H. SINGH

Group: A-A

YR 0191



Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	PP-QIS2-PV	Rev. 1	Page	16 of 23
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Attribute Data Sheet

Sample	Pass	Fail	Comments
1	X		
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X		
19	X		
20	X		
21	X		
22	XX		
23	XX		
24	X		
25	X		
26			
27			
28			
29			
30			
31	X/13	X/21/101	
32	X/2		
33			
34			
35			
36			
37			
38			
39			
40			

Date: 9-21-01

Performed By: H.SINGH

Group: A-B

VR 0191



N/A on 12/12/01

Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc No.	VP-0157.PV	Rev. I	Page	16 of 23
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Attribute Data Sheet

Sample	Pass	Fail	Comments
1	X		
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X		
19	X		
20	X		
21	X		
22	X		
23	X		
24	X		
25	X		
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39			
40			

Date: 9-21-01

Performed By: H-SIN GH

Group: A-C

W/H Mar 12/2001

VR 0191



Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev. 1	Page	16 of 23
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Attribute Data Sheet

Sample	Pass	Fail	Comments
1	X		
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X	N/A	11/29/2011
19	X		
20	X		
21	X		
22	X		
23	X		
24	X		
25	X		
26			
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28			
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39			
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Date: 9-28-01

Performed By: H SINGH

Group: B-A

VR 0191



2/12 MUL/2011

Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev.	I	Page	15 of 23
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Attribute Data Sheet

Sample	Pass	Fail	Comments
1	X		
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X		
19	X		✓/A
20	X		HS 9/20/01
21	X		
22	X		
23	X		
24	X		
25	X		
26			
27			
28			
29			
30			
31			
32			
33			
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35			
36			
37			
38			
39			
40			

Date: 9-28-01

Performed By: H SINGH

Group: B-B

VR 0191



Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev. 1	Page	15 of 23
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Attribute Data Sheet

Sample	Pass	Fail	Comments
1	X		
2	X		
3	X		
4	X		
5	X		
6	X		
7	X		
8	X		
9	X		
10	X		
11	X		
12	X		
13	X		
14	X		
15	X		
16	X		
17	X		
18	X		
19	X		A
20	X		H/S
21	X		
22	X		9/28/01
23	X		
24	X		
25	X		
26			
27			
28			
29			
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38			
39			
40			

Date: 9-28-01Performed By: H SINGHGroup: B-C

VR 0191

1/14 12/12/01

Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev.	1	Page	15 of 23
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BPGS
SFO 650
INV 180

Outer Coil Subassembly
QAF-1701, Lot History Record for 100795 (E0644-02)

Shop Order

Batch Number

Lot Number

LOT C

Process	Material	Lot No.	Initial	Date	Gly. Passed
1. Equipment Setup	E0578-02 Mandrel Bead 100936	Lot no. _____	_____	_____	N/A
2. Line Clearance SOP00922 rev()	N/A	N/A	_____	_____	N/A
3. QA Verification of Line Clearance SOP00922 rev()	N/A	N/A	_____	_____	N/A
4. Wind, Anneal R1526 rev(F) E1524 rev(J)	E0482-04 NTI Ribbon 100776	Lot no. <u>0102 0000 46</u> <u>PERFORMED BY ROSS</u>	RC	9/5/01	N/A
5. Trim, Inspect Q1524 rev(C) VS1554 rev(C)	Inspection Notes: <u>Spool B</u>	_____	RC	9/5/01	_____
6. Etch Coils R1727 rev() R0704 rev()	Hydrogen Peroxide 100810 70% IPA 100177 Etching Concentrate 100909 Deionized Water 100798 Alconox Powder 100762 Sodium Bicarbonate 100932	Lot no. <u>0106 0000 67</u> exp _____ Lot no. <u>0105 0000 57</u> exp 11/11/01 Lot no. <u>0106 0000 48</u> exp 6/11/01 Lot no. <u>0109 0000 79</u> exp 10/31/01 Lot no. <u>0100 0000 58</u> Lot no. <u>0105 0000 19</u>	RC C.S. mm CP	9/11/01 9/11/01 9/11/01	_____
7. Inspect Etching Q1718 rev()	Inspection Notes: <u>OK 9/11/01</u>	_____	_____	_____	_____
8. Make Bump R1601 rev()	E0700-05 Wire for Bump 100816	<u>OK</u>	_____	_____	as needed
9. Weld Bump R0204 rev()	E0503-11 Band, Platinum 100777	Lot no. <u>0108 0000 62</u>	RC LA.	9/13/01	_____
10. Stage Sandwich R0702 rev(E)	E0503-12 Ring, Platinum 100814	Lot no. <u>0108 0000 14</u>	RC LA.	9/13/01	_____
11. Weld Sandwich R0702 rev(E)	N/A	N/A	LA.	9/13/01	_____
12. Inspect Outer Coil E0644-02 rev() Q0644 rev()	Inspection Notes:	_____	_____	_____	_____
13. Cycle Test R0735 rev()	Inspection Notes: <u>OK 9/11/01</u>	_____	_____	_____	_____
14. Outer Coils removed for additional testing		Gly removed for testing	_____	_____	Gly removed

Additional Comments: (Record step number if comments are related to a particular step.)

VR 0191

Revision E	Conceptus CONFIDENTIAL Document QAF-1701 LHR for 100795 (E0644-02) - Outer Coil Subassembly	Page 1 of 1
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Conceptus

BPCS
SFC 650
INV 130

Outer Coil Subassembly QAF-1701, Lot History Record for 100795 (E0644-02)

Shop Order

Batch Number

Lot Number

LOT R

Process	Material	Lot No.	Initial Date	Qty. Processed
1. Equipment Setup	E0678-02 Mandrel Bead 100938	Lot no. _____	_____	N/A
2. Line Clearance SOP00922 rev()	N/A	N/A	_____	N/A
3. QA Verification of Line Clearance SOP00922 rev()	N/A <i>Appendix</i>	N/A	_____	N/A
4. Wind, Anneal R1525 rev() E1524 rev()	E0482-04 NTI Ribbon 100776	Lot no. <u>0103000045</u> <i>PERFORMED BY Ross RC</i>	mm 9/3/01 9/3/01	100% N/A
5. Trim, Inspect Q1524 rev() VS1524 rev()	Inspection Notes:	_____	RC 9/3/01	_____
6. Etch Coils R1727 rev() R0704 rev()	Hydrogen Peroxide 100810 70% IPA 100177 Etching Concentrate 100809 Deionized Water 100798 Alconox Powder 100762 Sodium Bicarbonate 100932	Lot no. 0106000067 exp 1/31/05 Lot no. 0105000057 exp 1/31/04 Lot no. 0106000049 exp 6/1/02 Lot no. 0108000179 exp 10/31/02 Lot no. 0105000058 Lot no. 0105000019	RC C.S. CR mm 9/1/01 9/1/01 9/1/01 9/1/01	_____
7. Inspect Etching Q1718 rev()	Inspection Notes:	_____	_____	_____
8. Make Bump R1501 rev()	E0700-05 Wire for Bump 100815	_____	mm 9/13/01	As needed
9. Weld Bump R0701 rev()	E0503-17 Band, Platinum 100777	Lot no. <u>0108000062</u> <i>PERFORMED BY Ross RC</i>	mm 9/13/01 9/13/01	100% N/A
10. Stage Sandwich R0702 rev()	E0503-12 Ring, Platinum 100814	Lot no. <u>0108000014</u>	mm 9/13/01 9/13/01	100% N/A
11. Weld Sandwich R0702 rev()	N/A	N/A	mm 9/13/01	100% N/A
12. Inspect Outer Coil E0644-02 rev() Q0844 rev()	Inspection Notes:	_____	_____	_____
13. Cycle Test R0735 rev()	Inspection Notes:	mm 9/4/01	_____	_____
14. Outer Coils removed for additional testing	_____	Qty removed for testing	_____	Qty remaining <i>See Control</i>

Additional Comments: (Record step number if comments are related to a particular step.)

VR 0191

Revision E	Conceptus CONFIDENTIAL Document QAF-1701 LHR for 100795 (E0644-02) - Outer Coil Subassembly	Page 1 of 1
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BPOS
SFC 650
INV 130

Outer Coil Subassembly
QAF-1701, Lot History Record for 100795 (E0644-02)

Shop Order _____ Batch Number _____ Lot Number Lot A

Process	Material	Lot No.	Initial	Date	Qty Passed
1. Equipment Setup	E0678-02 Mandrel Bead 100938	Lot no. _____			N/A
2. Line Clearance SOP00922 rev()	N/A	N/A			N/A
3. QA Verification of Line Clearance SOP00922 rev()	N/A	N/A			N/A
4. Wind, Anneal R1526 rev(F) E1524 rev(J) ^{9/5/01}	E0492-04 NITI Ribbon 100776	Lot no. <u>Q102 000046</u> PERFORMED BY ROSS	mu RC	9/5/01 9/5/01	N/A
5. Trim, Inspect Q1524 rev(K) VS1854 rev(E) ^{9/5/01}	Inspection Notes: <u>SPool 14</u>		RC	9/5/01	
6. Etch Coils R1727 rev() R0704 rev()	Hydrogen Peroxide 100810 70% IPA 100177 Etching Concentrate 100809 Deionized Water 100798 Alconox Powder 100762 Sodium Bicarbonate 100832	Lot no. <u>Q106 000067</u> exp 1/31/03 Lot no. <u>Q105 000057</u> exp 1/31/04 Lot no. <u>Q106 000049</u> exp 6/1/02 Lot no. <u>Q107 000125</u> exp 10/31/02 Lot no. <u>Q105 000058</u> Lot no. <u>Q105 000019</u>	CR as. as. as. as. as. CS RC	9/11/01 9/11/01 9/11/01 9/11/01 9/11/01 9/11/01	
7. Inspect Etching Q1718 rev()	Inspection Notes: <u>Pass 9/11/01</u>				
8. Make Bump R1501 rev()	E0700-05 Wire for Bump 100755	<u>Not yet started.</u>			as needed
9. Weld Bump R0703 rev()	E0503-11 Band, Platinum 100777	Lot no. <u>Q108 0000267</u>	LC IA.	9/13/01	
10. Stage Sandwich R0702 rev(E)	E0503-12 Ring, Platinum 100814	Lot no. <u>Q108 0000147</u>	LC IA.	9/13/01	
11. Weld Sandwich R0702 rev(E)	N/A	N/A	IA.	9/13/01	
12. Inspect Outer Coll E0844-02 rev() Q0544 rev()	Inspection Notes:				
13. Cycle Test R0735 rev()	Inspection Notes: <u>Pass 9/11/01</u>				
14. Outer Coils removed for additional testing		Qty removed for testing.			Qty remaining

Additional Comments: (Record step number if comments are related to a particular step.)

NR 0191



Revision E	Conceptus CONFIDENTIAL Document QAF-1701 LHR for 100795 (E0644-02) - Outer Coil Subassembly	Page 1 of 1
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WORK INSTRUCTION ACID PREPARATION

R1727

Revision Level	ECO#	Effective Date	Originators Name/ Signature/Date	Change Request
A	3641	07/18/01	M. Martinez	Initial Release. Acid Solution preparation, removed from R0704.
B	3783	08/21/01	M. Martinez	Add using pyrex dish for solution prep.

*Redlined
Document
M. Martinez 12/12/01.
J.*

VR 0191



Revision Level B	Conceptus CONFIDENTIAL Document R1727 Work Instruction Acid Preparation	Page 1 of 4
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1.0 PURPOSE:

This procedure defines the work instruction for preparing the acid solution for etching the annealed outer coil.

2.0 DOCUMENTS:

2.1 As specified on device specific DMR.

3.0 PRODUCTION PARTS:

3.1 As specified on device specific DMR.

4.0 EQUIPMENT

4.1 Hot Plate with magnetic stirring mechanism	FA90604 or FA90677
4.2 Ultrasonic Cleaner	

5.0 MATERIALS

5.1 Detergent	E0071-01
5.2 De-ionized Water	E0075
5.3 Etching Concentrate, ESTER-07	E0706
5.4 Hydrogen Peroxide, 30%	E0707
5.5 Isopropyl Alcohol, 70%	E0074
5.6 Sodium Bicarbonate	E1598

6.0 TOOLS/SUPPLIES

6.1 Latex or nitrile gloves	
6.2 Safety glasses or goggles	
6.3 Lab coat or chemical resistant apron	
6.4 Plastic, Graduated cylinder	
6.5 Syringe or pipette	
6.6 Wire basket for Ultrasonic Bath	
6.7 Tweezers	
6.8 Lint free wipes	
6.9 Vials	
6.10 Pyrex Dish 190x100 No. 3140	
6.11 Thermometer or Temperature Probe	FA90676
6.12 600g Scale	FA90642
6.13 1" Magnetic stirrer	
6.14 Glass, Graduated Cylinder 25 ml	
6.15 Glass, Graduated Cylinder 100ml	
6.16 Glass, Beaker 100 ml	
6.17 Glass, Beaker 1000 ml	
6.18 Glass, Beaker 400 ml	

7.0 Procedure

7.1 All glass ware is to be cleaned with detergent (E0071-01), prior to production.

7.2 Rinse glassware with De-ionized water (E0075).

7.3 Wear waterproof gloves, eye protection, and a lab coat or apron during solution preparation and use.

Y R 0197



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DEIONIZED WATER (E0075)

- 7.4 Place a dry, clean 100 (ml) graduated cylinder on the scale.
- 7.5 Zero out the scale.
- 7.6 Pour De-ionized water into the graduated cylinder until the digital readout is 181 ± 1 g.
- 7.7 Pour De-ionized water into a 1000 (ml) glass beaker or Pyrex Glass Dish 190x100 No. 3140.
- 7.8 Pour excess water into a waste container.

ETCHING CONCENTRATE (E0706)

- Note:** Keep etching concentrate at/or above room temperature.
- 7.9 Mix the Etching concentrate ESTER-07 (E0706) by swirling the contents.
 - 7.10 Place a dry, clean, 100 (ml) graduated cylinder on the scale.
 - 7.11 Zero out the scale.
 - 7.12 Pour etching concentrate into the graduated cylinder until the digital readout is 272 ± 1 (g).
 - 7.13 Pour etching concentrate inwith the De-ionized water.
 - 7.14 Pour excess solution into Acid waste Container.

HYDROGEN PEROXIDE (E0707)

- 7.15 Carefully loosen the cap of the HYDROGEN PEROXIDE, 30% (E0707).
- 7.16 Place a dry, clean 100 (ml) graduated cylinder on the scale.
- 7.17 Zero out the scale.
- 7.18 Pour hydrogen peroxide into the 100 (ml) beaker until the digital readout is 45 ± 1 (g).
- 7.19 Pour the hydrogen peroxide inwith De-ionized water and etching concentrate.
- 7.20 Pour excess peroxide into Acid Waste Container

ACID SOLUTION

- 7.21 Mix the etching solution with a stirring rod.
- 7.22 Pour 68ml \pm 3ml of the etching solution into the 100 ml graduated cylinder or 100 ml beaker.
- 7.23 Label 100 (ml) graduated cylinder Reserve Acid.
- 7.24 Pour the remaining solution into Pyrex Glass Dish 190x100 No. 3140.
- 7.25 Place-dish on hot plate.
- 7.26 Set the magnetic stirrer to 150 RPM.
- 7.27 Place 1" magnetic stirrer in acid solution.
- 7.28 ~~Place the temperature probe into acid solution~~
- 7.29 ~~Get hot plate temperature to $60^{\circ}\text{C} \pm 2^{\circ}\text{C}$~~
- 7.30 ~~Heat acid solution for approximately 30 minutes~~

W
4/4/01

VR 0191



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SODIUM BICARBONATE (E1598)

- 7.31 Mix $\frac{1}{2}$ a spoonful of bicarbonate in a beaker with 300 (ml) of De-ionized water.

9.0 INSPECTION CRITERIA:

- 31 N/A

10.0 - SHUT DOWN/CLEAN UP.

- 10.1 Turn off the hot plate.
 - 10.2 Turn off the ultra sonic cleaner.

11.0 TRAINING.

- 11.1** Training must be performed by a process expert.

11.2 Personnel being trained must be able to produce 2 complete batches of acid solution and accompanying solutions.

m k s

VR 0191



Revision Level · B	Conceptus CONFIDENTIAL Document R1727 Work Instruction Acid Preparation	Page 4 of 4
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WORK INSTRUCTION-OXIDE REMOVAL

R0704

Revision Level	ECO#	Effective Date	Originators Name/ Signature/Date	Change Request
A	2905	8/23/00	Andrew Ortega	Update with new equipment (scale, stir plate, glassware); changed solution measurement from milliliters to grams.
B	2966	9/27/00	Andrew Ortega	Update etching lapse time from 24 hrs to 7 days. Specified orientation of fixtures in beaker.
C	3152	12/14/00	Andrew Ortega	Changed Etching solution; increased temp from 30 – 40C to 60C; add neutralizer and 2 stage rinse; etch proximal end first.
D	3318	03/22/01	Andrew Ortega	Add neutralize and rinse during 10 minute etching time. Added one time 20 minute ultrasonic clean, was 30 min on each side separately.
E	3397	04/25/01	M. Martinez	Add Training, Inspection criteria, shut down. Clean document up, Update document to reference wound outer coil (E1524) instead of outer coil sub-assembly gamma device (E1524)
F	3427	05/02/01	M. Martinez	Increase ultrasonic rinse time in water for oxide removal. Add coil rework.
G	3640	07/18/01	M. Martinez	Rewrite document. Acid solution preparation removed to new Document R1727
H			M. Martinez	Add new process window and instructions per VP-0167.

MM
9/27/01

VR 0191



ReLine

Revision Level

6/14

of 9/27/01

Conceptus CONFIDENTIAL Document
R0704 Work Instruction-Oxide Removal

Page 1 of 6

1.0 PURPOSE:

This procedure defines the work instruction for removing the Titanium Oxide from the Annealed Outer Coil Surface. E1524

2.0 DOCUMENTS:

2.1 As specified on device specific DMR.

3.0 PRODUCTION PARTS:

3.1 As specified on device specific DMR.

4.0 EQUIPMENT

4.1	Ultrasonic Cleaner	
4.2	Etching Fixture	E0772
4.3	Pyrex Dish 190x100 No. 3140	(For etching solution)
4.4	Thermometer or Temperature Probe	FA90676
4.5	Hot Plate with magnetic stirring mechanism	FA90604 or FA90677
4.6	1" Magnetic stirrer	
4.7	Glass, Graduated Cylinder 25 ml	
4.8	Glass, Graduated Cylinder 100ml	
4.9	Glass, Beaker 100 ml	
4.10	Glass, Beaker 1000 ml	
4.11	Glass, Beaker 400 ml	

5.0 MATERIALS

5.1	De-Ionized Water	E0075
5.2	Etching Concentrate, ETSER-07	E0706
5.3	Hydrogen Peroxide, 30%	E0707
5.4	Isopropyl Alcohol, 70%	E0074
5.5	Sodium Bicarbonate	E1598

6.0 TOOLS/SUPPLIES

6.1	Water proof gloves
6.2	Safety glasses or goggles
6.3	Lab coat or chemical resistant apron
6.4	Plastic, Graduated cylinder
6.5	Syringe or pipette
6.6	Wire basket for Ultrasonic Bath
6.7	Tweezers
6.8	Lint free wipes
6.9	Vials

8.0 PROCEDURE**Fixture Preparation (E0772):**

NOTE: The subassembly quantity is limited to a maximum of 25 assemblies per fixture. A maximum of 5 fixtures may be used in the solution at any given time.

VR 0191



Revision Level H	Conceptus CONFIDENTIAL Document R0704 Work Instruction-Oxide Removal	Page 2 of 6
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- 8.1 Position the proximal end of the outer coil subassemblies on the same side of etching fixture.
- 8.2 Verify that the proximal tip of the outer coil is touching the end of the fixture.
- 8.3 Position Delrin clip between the press to fit pins.
- 8.4 Press clip down.

CLEAN COILS 70% IPA (E0074):

- 8.5 Transfer 400 (ml) of 70% IPA into a 1000 (ml) beaker.
- 8.6 Place fixtures into beaker.
- 8.7 Orient the fixture(s) so that the subassemblies are horizontal in beaker.
- 8.8 Place Beaker in ultra sonic cleaner.
- 8.9 Set timer to 5 minutes.
- 8.10 Start timer.
- 8.11 Remove the fixture(s) and subassemblies from the Isopropyl Alcohol.
- 8.12 Allow components to air-dry for a minimum of 2 minutes.
- 8.13 Remove clip from the etching fixture and reposition the tip of the coils to touch fixture.
- 8.14 Reposition clip.
- 8.15 Using a microscope verify that the coils are straight and touch the tip of the etching fixture.

ETCHING:

Proximal Tip (short side):

- 8.16 Set timer to 22'(min.) 30"(sec.).
- 8.17 Place fixtures in etching solution.
- 8.18 Start timer.
- 8.19 Dip plastic ruler into solution.
- 8.20 Verify that the solution level is within $1.6 \pm .1$ (cm).
- 8.21 If required fill solution level to specified height.
- 8.22 Remove fixtures after timer has signaled the end of process.
- 8.23 Submerge entire fixture in neutralizer for a minimum of 15 seconds.
- 8.24 Submerge entire fixture in beaker number 1 (300 (ml) De-ionized water), for a minimum of 15 seconds.
- 8.25 Submerge entire fixture in beaker number 2 (300 (ml) De-ionized water), for a minimum of 15 seconds.
- 8.26 Remove clip from the etching fixture and reposition the tip of the coils to touch fixture.
- 8.27 Reposition clip.
- 8.28 Using a microscope verify that the coils are straight and touch the tip of the etching fixture.
- 8.29 Set timer to 22'(min.) 30"(sec.).
- 8.30 Place fixtures in etching solution.

Y R 0191



Revision Level H	Conceptus CONFIDENTIAL Document R0704 Work Instruction-Oxide Removal	Page 3 of 6
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- 8.31 Start timer.
 - 8.32 Dip plastic ruler into solution.
 - 8.33 Verify that the solution level is within $1.5 \pm .1$ (cm).
 - 8.34 If required fill solution level to specified height.
 - 8.35 Remove fixtures after timer has signaled the end of process.
 - 8.36 Submerge entire fixture in neutralizer for a minimum of 15 seconds.
 - 8.37 Submerge entire fixture in beaker number 1 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.38 Submerge entire fixture in beaker number 2 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.39 Remove fixtures from the etching solution.
- Distal Tip (Long Side):**
- 8.40 Remove the clip from the etching fixture and position the Distal tip of the coils to touch fixture.
 - 8.41 Reposition clip.
 - 8.42 Using a microscope verify that the coils are straight and touch the tip of the etching fixture.
 - 8.43 If volume is below $1.5 \pm .1$ (cm), refill to specified level.
 - 8.44 Set timer to 22'(min.) 30"(sec.).
 - 8.45 Place fixtures in etching solution.
 - 8.46 Start timer.
 - 8.47 Dip plastic ruler into solution.
 - 8.48 Verify that the solution level is within $1.5 \pm .1$ (cm).
 - 8.49 If required fill solution level to specified height.
 - 8.50 Remove fixtures after timer has signaled the end of process.
 - 8.51 Submerge entire fixture in neutralizer for a minimum of 15 seconds.
 - 8.52 Submerge entire fixture in beaker number 1 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.53 Submerge entire fixture in beaker number 2 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.54 Remove clip from the etching fixture and reposition the tip of the coils to touch fixture.
 - 8.55 Reposition clip.
 - 8.56 Using a microscope verify that the coils are straight and touch the tip of the etching fixture.
 - 8.57 Set timer to 22'(min.) 30"(sec.).
 - 8.58 Place fixtures in etching solution.
 - 8.59 Start timer.
 - 8.60 Dip plastic ruler into solution.
 - 8.61 Verify that the solution level is within $1.5 \pm .1$ (cm).
 - 8.62 If required fill solution level to specified height.
 - 8.63 Remove fixtures after timer has signaled the end of process.



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Revision Level H	Conceptus CONFIDENTIAL Document R0704 Work Instruction-Oxide Removal	Page 4 of 6
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- TA Marseille 2009151 - reçu le 13 janvier 2021 à 16:14 (date et heure de métropole)
- 8.64 Submerge entire fixture in neutralizer for a minimum of 15 seconds.
 - 8.65 Submerge entire fixture in beaker number 1 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.66 Submerge entire fixture in beaker number 2 (300 (ml) De-ionized water), for a minimum of 15 seconds.
 - 8.67 Remove fixtures from the etching solution.

ULTRA-SONIC CLEANING DI WATER

- 8.68 Set timer to a minimum of 90 minutes.
- 8.69 Place fixtures in the ultra sonic water bath.
NOTE: Use fresh water in the ultra sonic cleaner.
- 8.70 Transfer the fixtures to 400ml beaker(s) filled with de-ionized water (E0076).
- 8.71 Fill the ultrasonic cleaner to the operational level with water as indicated on the bath.
- 8.72 Place beaker(s) into the wire mesh basket of the ultrasonic cleaner.
- 8.73 Start timer.
- 8.74 Remove fixtures after alarm has signaled.

ULTRA-SONIC CLEANING 70% IPA

- 8.75 Transfer fixtures to 400 (ml) beaker(s) of 70% IPA.
- 8.76 Set timer for 2 minutes.
- 8.77 Orient the fixture(s) so that the subassemblies are horizontal in beaker.
- 8.78 Place beakers in ultra sonic cleaner.
- 8.79 Start Timer.
- 8.80 Remove fixtures from IPA after alarm signals.
- 8.81 Place device components on lint free cloth.
- 8.82 Allow components to dry for a minimum of 2 minutes.
- 8.83 Place the outer coils into plastic vials.
- 8.84 Record the time and date etching was completed on the build document.

WASTE DISPOSAL

- 8.85 Dispose of all solution after one complete cycle.
- 8.86 Pour waste solution in ETSER-07 Waste Container.
- 8.87 Pour waste de-ionized water in sink.
- 8.88 Pour waste IPA in Waste Container.
- 8.89 Clean all equipment with detergent.
- 8.90 Rinse and allow to air dry.

VR 0191



Revision Level H	Conceptus CONFIDENTIAL Document R0704 Work Instruction-Oxide Removal	Page 5 of 6
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9.0 INSPECTION CRITERIA:

- 9.1 Inspect both distal and proximal ends of outer coils for etching. Reject if the coil is not etched a minimum of 1 mm.
- 9.2 There must be no oxide either proximal or distal located along the first mm of the coil.
- 9.3 Light oxide may be present from 1 mm UP TO SEVEN COILS.

10.0 SHUT DOWN:

- 10.1 Turn off the hot plate.
- 10.2 Turn off the ultra sonic cleaner.
- 10.3 Remove water form ultra sonic cleaner.

11.0 TRAINING:

- 11.1 Training must be performed by a process expert.
- 11.2 Personnel being trained must produce 50 etched outer coils before being allowed to manufacture production assemblies.
- 11.3 Units assembled for training, must be inspected by the designated process expert and pass the specification.

VR 0191



Revision Level H	Conceptus CONFIDENTIAL Document R0704 Work Instruction-Oxide Removal	Page 6 of 6
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INSPECTION INSTRUCTIONS, ETCHING

1.0 TOOLS AND EQUIPMENT

- 1.1 Microscope
- 1.2 Tweezers

2.0 INSPECTION:

NOTE: INSPECTION TO BE PERFORMED USING A 10X MICROSCOPE

- 4.1 Inspect the proximal and distal tip of the outer coil, all oxide and flakes must be completely removed a minimum of 1mm, see Figure 1.
- 4.2 One millimeter from the proximal and distal tip, etching is acceptable within the first 7 coils, see Figure 1 and Figure 4.
- 4.3 Silver streaks located in the etched areas is acceptable, see Figure 2.
- 4.4 Spotting or longitudinal streaks located in the etched areas are acceptable, see Figure 1, 3, and 4.

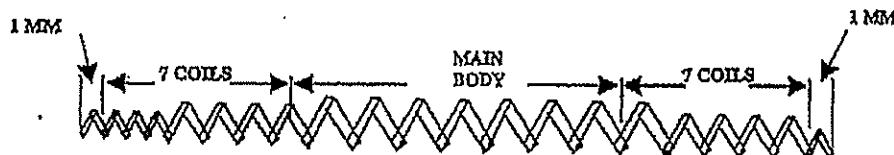


FIGURE 1 - DEFINITION OF PARTS OF THE COIL

1/10 mn 12/12/101

Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev.	1	Page	9 of 23
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VR 0191



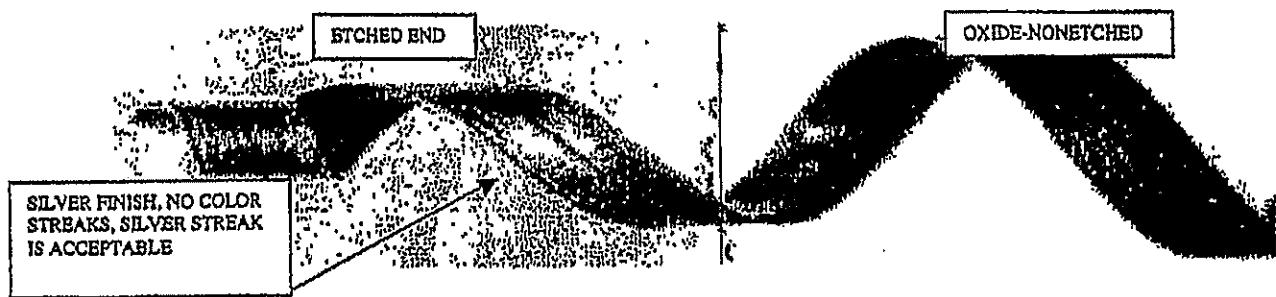


Figure 2- ETCHED PROXIMAL OUTER COIL.

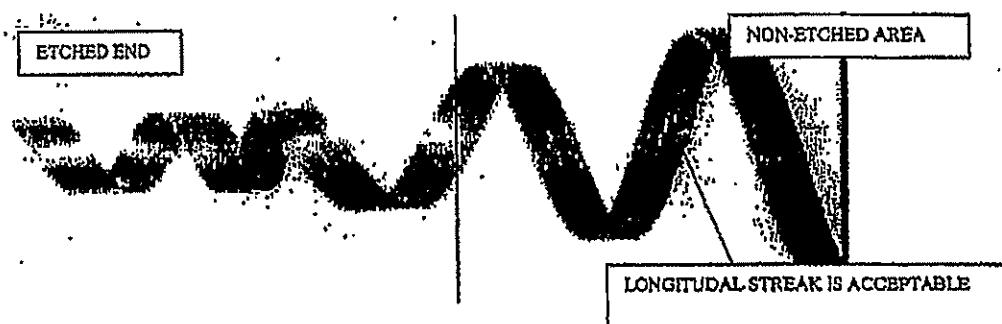


Figure 3- ETCHED DISTAL END OF OUTER COIL.

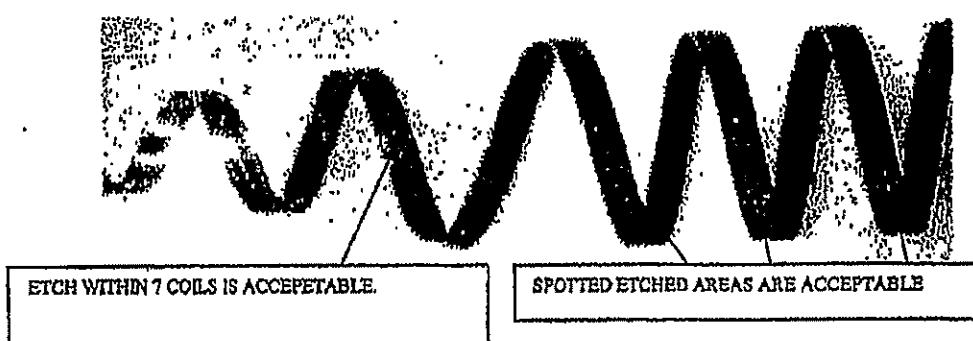


Figure 4 -LIGHT ETCH and SPOTTED ETCH IN THE MAIN BODY OF THE COIL.

Description	Outer Coil Oxide Removal Operation & Process Qualification Protocol	Doc. No.	VP-0157.PV	Rev.	1	Page	10 of 23
<i>MM 12/12/01</i>				VR	0191		



APPENDIX 4

64 PROJECT Corrosion Test - Water Bath Set-up Monitor (Continued From Page 1) Notebook No. 259

Corrosion Susceptibility Test Protocol on Concrete Specimen

(Spec. doc. 14 P0191 Rev 1)

Equipment used:

Water bath (VWR model 1240 S/N 1007388) This water bath
calibration date 3-10-02 calibration ID# 1100000197 not used
Date 11-19-01

Fluke 52-II Thermometer ID# FA90694

Calibration Date 3-12-02
3-10-02

Calibration Date 3-22-02 / 3-10-03

D.T. Labrad. Lot# 01100000197

Water Bath set-up: water bath VWR model 1240 S/N 1007388

10-18-01

Bath was filled with tap water to a depth of 1 1/4". Bath temperature was set to 37°C + covered. The temperature was monitored for 4 days.

10-12-01, 8:30 AM

The water bath was emptied & refilled with tap water to a depth of 1 1/4". The temperature was set at 32°C

At 12:30, the corrosion samples were placed in the bath, the temperature was 27°C at the bottom of the bath and 36.8°C

at the top of the water level. The temperature was taken in the center of the tank.

Solution Preparation:

Sodium Chloride 85-90% Lot# R0014005 Exp. Date 3-31-02

Buffer Solution Ph 7.00 Lot# 20100228

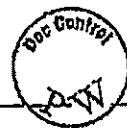
10-01-2001

Mixed 150 ml of Sodium Chloride + 2.0 drops of Buffer Solution in 100ml beaker. The pH was tested with indicators. Small amounts of buffer solution were added until the pH was between 6-7.0.

Continued on Page 65

Read and Understood By

Chris L.



Signed

11-19-01

Date

VR 0191

1-11-02

Date

d

PROJECT Coronation Test Water bath serial monitoring, Continued From Page .. 64 .. Notebook No. 239 65

Date	Time	Temp °C	Date	Time	Temp
10-8	8:30	26.2 / 26.1	11-10-02	21:30	32.6 / 32.8
10-8	2:30	32.6 / 32.6	11-12-02	10:45	32.8 / 32.6
10-9	21:30	32.8 / 32.5	11-23	9:30	32.2 / 32.3
10-10	9:00	36.5 / 36.4	1-30	8:00	32.3 / 32.4
10-13	3:20	36.1 / 36.1	2-6-02	11:30	36.5 / 36.7
10-11	8:00	36.9 / 32.0	2-13-02	8:30	36.8 / 36.9
10-11	4:30	37.4 / 32.4	2-20	9:30	32.1 / 32.2
10-12	12:50	32.0 / 36.8			
10-12	4:30	32.7 / 32.6			
10-15	8:00	36.4 / 36.9			
10-16	8:30	36.9 / 36.8			
10-17	8:00	36.9 / 36.9			
10-18	10:30	32.3 / 32.3			
10-19	11:30	36.8 / 36.8			
10-22	7:30	37.4 / 32.4			
10-23	9:00	32.1 / 32.2			
10-24	7:00	37.3 / 37.4			
10-25	2:15	36.9 / 32.1			
10-26	7:30	36.3 / 36.4			
10-29	7:00	36.1 / 36.9			
10-31	8:00	36.8 / 36.8			
11-2	8:00	32.2 / 32.1			
11-5	3:00	32.3 / 32.4			
11-7	10:00	36.4 / 32.0			
11-12	1:30	32.3 / 32.3			
11-20	9:30	32.0 / 32.3			
11-24	11:30	36.9 / 32.0			
12-5	2:30	32.3 / 32.4			
12-11	11:00	32.0 / 33.0			
12-18	7:00	32.5 / 32.6			
1-3-02	10:40	32.2 / 32.2			

Read and Understood By

Chayh

Signed

2-22-02

Date

Ehsan Alabdullah

Signed

2-22-02

Date



V R 0191

Continued on Page 66

Corrosion Susceptibility Test - Part of the Corrosion Project
Specified VPO Grade Level

Objective: to record vial numbers and dates that test strip
was started & completed.

Procedure in Lab notebook # 248 pg 7 was followed when
removing devices from the vials - group A, and when replacing
vials with solution - group B.

Oct. 17, 2001

Group A - All of group A vials with samples were placed in the
water bath - 2 vials total.

Group B vials numbered B-1-a through B-1-f were placed
in the water bath.

Oct. 19, 2001

Group A - vials numbered A-1-a, A-1-b, A-1-c were removed.

Group B - Devices from vials numbered B-1-a thru B-1-f were
removed & placed into vials with solution numbered
B-2-a thru B-2-f, then placed back in the water bath.



VR 0191

Continued on Page 67

Read and Understood By

Chad L.

Signed

10-19-01

Date

Elaine Aldridge

Signed

6-25-2002

Date

PROJECT Corrosion Test Vial numbers & Test Dates

Notebook No. 239

Continued From Page 66

67

Oct 26, 2001

Nov 11, 2001
as, 11-11-01

Group A - vials numbered A-2-a, A-2-b, A-2-c were removed.

Group B - Deutscher Form vials numbered B-2-a thru B-2-f were removed and placed into vials with solution numbered B-3-a thru B-3-f, then placed back in the water bath.

Nov 12, 2001

Group A - vials numbered A-3-a, A-3-b, A-3-c were removed

Group B - Deutscher Form vials numbered B-3-a thru B-3-f were removed and placed into vials with solution numbered B-4-a thru B-4-f, then placed back in the water bath.

Dec 11, 2001

Group A - vials numbered A-4-a, A-4-b, A-4-c were removed

Group B - Deutscher Form vials numbered B-4-a thru B-4-f were removed and placed into vials with solution numbered B-5-a thru B-5-f, then placed back in the water bath.

Jan 10, 2002

Group A - vials numbered A-5-a, A-5-b, A-5-c were removed

Group B - Deutscher Form vials numbered B-5-a thru B-5-f were removed and placed into vials with solution numbered B-6-a thru B-6-f, then placed back in the water bath.

Continued on Page 68

Read and Understood

VR 0797

Chyl J

Signed

1-10-02

Date

Claudia Alchado

6-25-2002

Date

Feb. 11, 2002

Group A + Mials numbered A-6-A, A-6-B, A-6-C were removed.

Group B - Deviates from vials numbered B-6-A thru B-6-F were removed, it placed into vials with soleinum numbered B-7-A thru B-7-F, then placed back into the water bath.

Water bath Manufacturing Table continued from pg 65

DATE TIME TEMP C

2-27-02 2:50 37.9/32.9

3-6-02 8:30 37.8/32.9

3-13-02 2:00 37.7/32.7

3-20-02 10:30 37.6/32.8

3-27-02 8:10 37.3/32.4

4-3-02 8:15 32.1/32.2

4-10-02 2:15 32.2/32.6

March 11

The Deviates were removed from the water bath and visually inspected without removing them from the glass vials. All deviates were intact, with all components held solder in their original position. The vials were placed back into the water bath.

VR 0191

Continued on Page 95

Read and Understood By

Chad L

Signed

4-12-02

Date

Ehsia Aldridge

Signed

4-25-2002

Date



Apel 100:

Corrosion A: Units numbered A-7-a, A-7-b, A-7-c were removed from the water bath. The fixtures were removed from the walls and ultrasonically cleaned in DI water for 2 minutes.

Corrosion B: Units numbered B-7-a thru B-7-f were removed from the water bath. The fixtures were removed from the walls and ultrasonically cleaned in DI water for 2 minutes.

2.4g 6/25/02



VR 0197

Continued on Page 96

Read and Understood By

Signed

4-12-02

Date

Signed

16-25-2002

Date

PROJECT Corrosion Test U, A1 Numbers + Test Dates

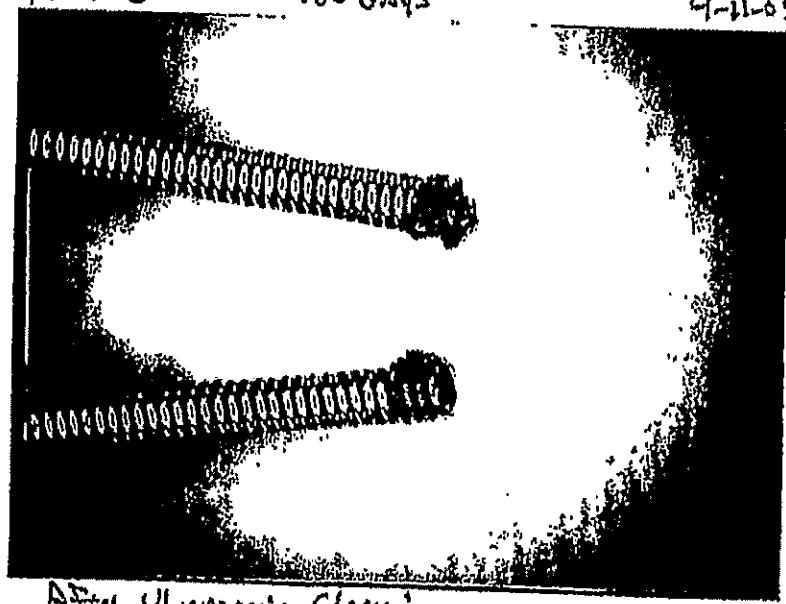
Notebook No. 239:
Continued From Page 95

After Ultrasonic Cleaning 6/25/2002

B-2-B

180 DAYS

4-11-02



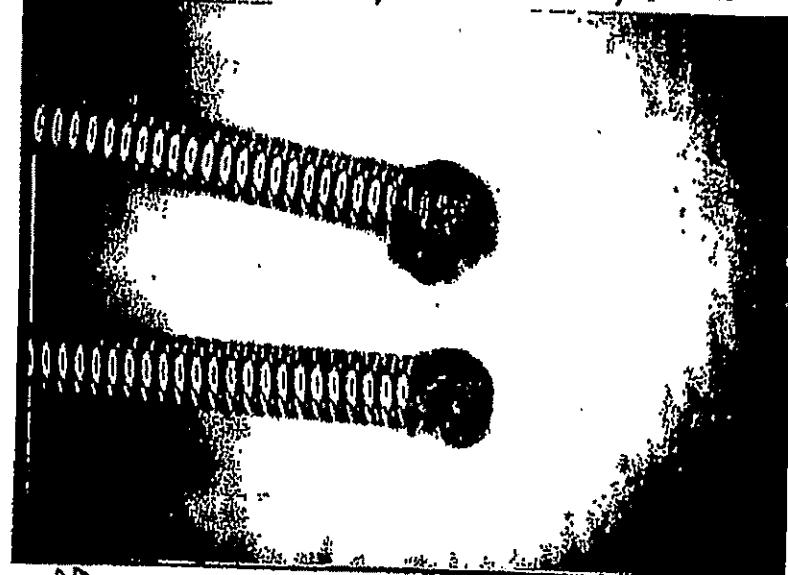
6/25/02

After Ultrasonic Cleaning

B-2-C

180 DAYS

4-11-02



6/25/02

After Ultrasonic Cleaning

VR 0791

Continued on Page 97

Read and Understood By

Cheryl L.

Signed

4-12-02
Date

Elisa Aldridge

Signed

4-25-2002
Date

PROJECT Corrosion Test Vial Numbers + Test Dates

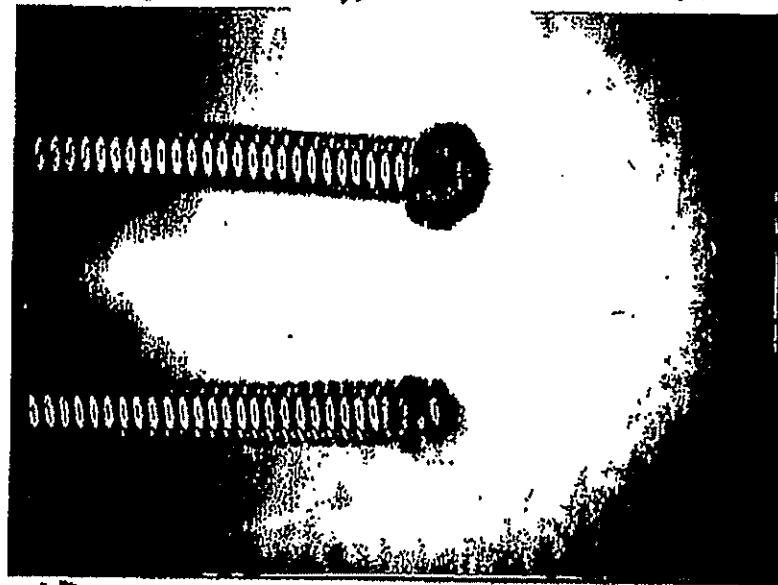
Notebook No. 239

Continued From Page 26

9:

B-7-D 180 DAYS

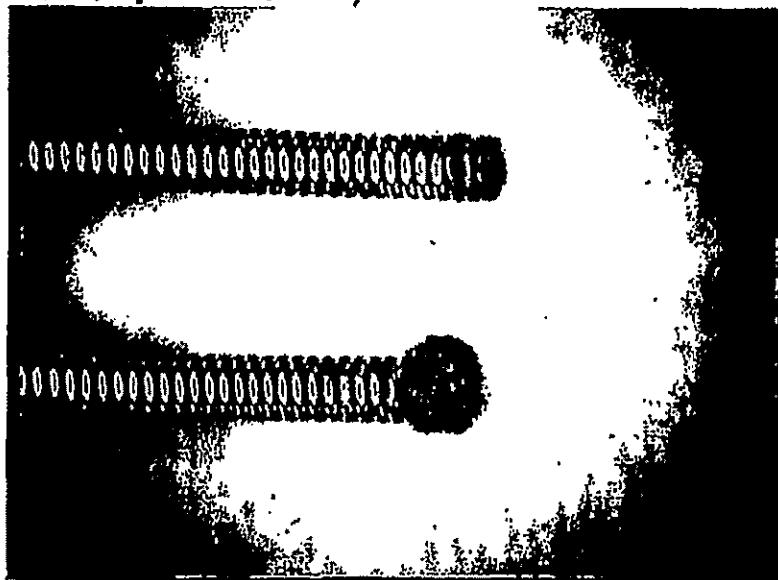
4-11-02



After Ultrasonic Cleaning

B-7-F 180 DAYS

4-11-02



After Ultrasonic Cleaning

VR 0191

Continued on Page _____

Read and Understood By

Chayl L

Signed

4-12-02

Date

Elsia Aldejga

Signed

6-25-2002

Date



24/02/02
25/02/02
6/12/02
1/13/02

PROJECT Corrosion test preparation (VR0191_pv rev1)

Notebook No. 248

Continued From Page

67

Objective:

To document the preparation of solutions, samples and water bath for VR0191_pv rev1. All documented steps were performed on 10/22/01.

Tools, equipment:

Bowl, capacity 250 ml, graduated, line at 150 ml = 5%.

Stainless steel flasks

Temperature Gauge Part # FTA90694 cal: 3/22/01 Due 3/22/02

Water bath, thermostatically controlled

Disposable Pipettes, VWR part # 141670-330, cap. 5 ml

pH paper:

colorpHast pH 2-9 Cal. 95-98 lot# 00458060

colorpHast pH 6.5-10.0 Cal. 95-93 lot# 00451990

colorpHast pH 4.0-9.0 Cal. 95-92 lot# 00451992

colorpHast pH 0-14.0 Cal. 95-90 lot# 00458453

colorpHast pH 7.5-14.0 Cal. 95-87 lot# 00458451

Vials

Materials:

Sodium Chloride 0.85 - 0.90 % solution w/w Lot # RD 0128405
Part # E1820 exp: 5-31-02

Buffer Solution pH 7.0 Lot # RD 0128401 exp: 3-31-03
Part # E1879-04

Device Subassembly Lot # RD 0129001 Sq # 1058
Part # E0648-02 rev C



VR 0191

Continued on Page 68

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[Signature]
Signed

11/5/01

Date

[Signature]
Signed

1/10/02
Date

68

PROJECT Corrosion test preparation VPO191, pr rev1

Notebook No. 248

Continued From Page

67

Solution preparation:

- Pour into 250 ml Beaker 160ml of Sodium Chloride solution E1840.
- Add 30 drops of pH 7.0 Buffer solution E1879-04.
- Verify that pH is in the 6.0 - 7.0 range.

Sample preparation on 19/12/01

Location : Concepts R/A lab

Present : Jay Yang, Christian Lowe, Cheryl Sorenson

A) Solution was prepared per the above formula pH was 6.5 ± 1/4 pH units measured with pH papers listed under materials

B) Vials are labeled per VPO191, pr rev1 Step 6

C) Vials from A series and B-1-a through B-1-f are filled with
solution from a) and 2 drops were added to each vial, cap was sealed.

D) Three control vials labeled Control 1-C, Control 2-C, Control 3-C were filled with

Control 1-C
Control 2-C
Control 3-C

Solution and sealed with cap.

E) Three vials were labeled C-A. No dropper was added to each
vial and cap was sealed.

C-A
C-B
C-C



F) Vials labeled B-2-a through B-2-f were filled with solution and
Cap was sealed, in preparation for day 7, Step 16 of VPO191, pr rev1.

P.S: Vials were filled to following level:
Right below Head

Sub-level

V

R

O

I

9

1

Continued on Page 69

Read and Understood By

Signed

11/15/01

Date

Signed

1/10/02

Date

PROJECT Codexian test preparation VP0191 per 2021

Notebook No. 248
Continued From Page 68

69

Vial disposition in racks:

- Two racks A) Series from VP0191 per 2021
B) B series from VP0191 per 2021

Schematic of disposer:

Rack A

A-1-C	A-2-C	A-3-C	A-4-C	A-5-C	A-6-C	A-7-C
A-1-B	A-2-B	A-3-B	A-4-B	A-5-B	A-6-B	A-7-B
A-1-A	A-2-A	A-3-A	A-4-A	A-5-A	A-6-A	A-7-A

1 2 3 4 5 6 7 P 9 10

Rack B

B-1-A	B-2-B	B-1-C	B-1-D	B-1-E	B-1-F

1 2 3 4 5 6 7 P 9 10



VR 0191

Continued on Page 70

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1/15/21

Date

Signed

1/15/21
Date

70
PROJECT Corrosion test preparation VPO191.pv.nwl

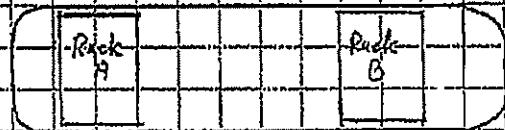
Notebook No. 248

Continued From Page 63

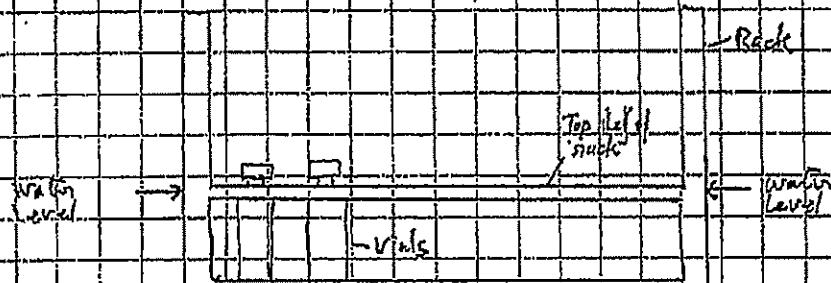
Water Bath =

Rack disposition in bath (Top view)

Front of Bath
control



Water Level in bath - side view of rack



Racks were placed in water bath on 10/12/01 at 12:30 pm



V R 0191

Continued on Page N/A
AKG/25/01

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11/5/01

Date

Signed

1/6/02
Date

Objective: To describe sample manipulation steps during UP0191.PV rev1

Group A:

1. Remove vials from sack
2. Using tweezers previously cleaned with alcohol remove device from vial
3. Clean devices by immersing in di water in ultrasonic cleaner for 1 min. Rinse dry device. Put into clean vials labeled with same code as the vial they came from.
4. Keep all original solutions in the original vials, screw cap on tightly.

2 min
10/10/02

Group B:

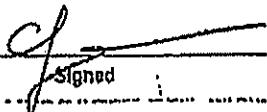
1. remove vials from sack
2. Using previously cleaned tweezers, remove devices from vial and place into appropriate distribution vial (per UP0191.PV rev1) Wear gloves during manipulation, blot gently dry before putting device into distribution vials. Close distribution vials, tighten cap, replace in sack.
3. Keep solutions ^{separately} in original vials, tighten down cap. Share in small places.



VR 0191

Continued on Page

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11/9/01

Date


Signed

1/10/02

Date

72

PROJECT Delivery of Solutions to analysis : VP0191.VP rev1 Notebook No. 248
Continued From Page

Scope: To record the delivery of solutions to analysis laboratory.

Location/lab information: Scientific Experimental Labs
Palo Alto, CA

Samples delivered to T. Yang on 11/14/01 for hand delivery to lab

Constituents: A = 1-A. 7 day Solution Controls:

A-1-B 7 day

c-d control

A-1-C 7 day

c-e control 2

A-2-B 14 day

c-f control 3

A-2-C 14 day

A-3-A 30 day

A-3-B 30 day

A-3-C 30 day

With change of solution:

B-1-A B-2-A B-3-A

B-1-B B-2-B B-3-B

B-1-C B-2-C B-3-C

B-1-D B-2-D B-3-D

B-1-E B-2-E B-3-E

B-1-F B-2-F B-3-F



VR 0191

Continued on Page

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Signer 1

11/14/01

Date

Signed

11/16/01

Date

PROJECT SEM Examination of core samples VP0191.1b rev1

Notebook No. 248

73

Continued From Page

Scope: Per VP 0191.1b rev1, samples from Corrosion test will be analyzed for evidence of corrosion by SEM examination.

Laboratory: Charles Evans Assoc., Wed 11/14/01, Analyst: C., Jay Young

Samples

Controls (0 day)	C-A C-B C-C	Cumulative A-1-A A-1-B A-1-C A-2-A A-2-B A-2-C A-3-A A-3-B A-3-C	7 day 3 day 3 day 14 days 14 days 14 days 30 days 30 days 30 days
---------------------	-------------------	---	---

SEM Model: Ted ISM 5800 LV

Operator: Han Wei Chen

Special sample prep: N/A

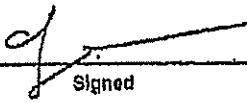
Layout:



VR 0791

Continued on Page 24

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11/14/01

Date


Signed

11/14/01
Date

ATA Marseille 2009151 - reçu le 13 janvier 2021 à 16:14 (date et heure de métropole)

Date / /

W. G. Miller
Signed

Date
1/09/02

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Read and Understood By

Continued on Page 3

WR-019-1

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A. a. C. Q 8 " " A. a. C.
Punjab : U. N. N. T. Y. D.
SPPU Mysore : Tech. Inst. 580081
Laycock and Draycot in SEIN

PROJECT CPM. Description of contract - VPO1911.P
Continued from Page 94
Notebooks No. 2

PROJECT SEM Examination VPA151.pr.nrl

Notebook No. 248

Continued From Page 76

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A4C01	Bell tip	35X
A4C02	"	1000X
A4C03	wire	150X
A4C04	"	1000X
A4C05	distal end of 1/2 solder	150X
A4C06	" "	1000X
A4C07	inner coil stopper / solder	150X
A4C08	" " "	1000X
A4C09	Cutter coil	150X
A4C10	Cutter coil	1000X
A4C11	cutter coil / band	150
A4C12	" "	1000X
A4A01	1/2 band at wire ribbon	150X
A4A02		1000X

Continued on Page



VR 0191

Read and Understood By

Signed

1/19/02

Date

Signed

Date

70
PT

CORROSION STUDY 90 DAY SEM PHOTOS

271

N/A

SEM TECHNICIAN: JANE WHEELER
 CHARLES EVANS & ASSOCIATES
 ACCOMPANIED BY CONSULTANT, JAY XANG
 PHOTOS TAKEN WED. 1-23-02.
 PHOTOS ARE NOT INCLUDED IN THIS NOTEBOOK.

<u>SAMPLE</u>	<u>PHOTO LABEL</u>	<u>DESCRIPTION</u>
A-5-a	E5417 A-5-a-1	Ball tip, 100x
	E5417 A-5-a-2	Ball tip, 2000x
	E5417 A-5-a-3	Inner coil 4-5 coils distal to solder bond, 100x
	E5417 A-5-a-4	Inner coil, 2000x
	E5417 A-5-a-5	Solder bond close to proximal end, 100x
	E5417 A-5-a-6	Solder bond " , 2000x
	E5417 A-5-a-7	Nitinol ribbon very close to solder, 2000x
	E5417 A-5-a-8	Nitinol ribbon in middle of O.C., 100x
	E5417 A-5-a-9	Nitinol ribbon in middle of O.C., 2000x
	E5417 A-5-a-10	Same as A-5-a-9, but darker
	E5417 A-5-a-11	Sandwich weld, 100x
	E5417 A-5-a-12	Sandwich weld, 2000x
	E5417 A-5-a-13	Solder at $\frac{1}{4}$ turn, 100x
	E5417 A-5-a-14	Solder at $\frac{1}{4}$ turn, 2000x
A-5-b	E5417 A-5-b-1	Ball tip, 100x
	E5417 A-5-b-2	Ball tip, 2000x
	E5417 A-5-b-3	Inner coil 4-5 coils distal to solder bond, 100x
	E5417 A-5-b-4	Inner coil " , 2000x
	E5417 A-5-b-5	Solder bond, prox. end, 100x
	E5417 A-5-b-6	Solder bond, solder to inner coil transition, 2000x
	E5417 A-5-b-7	Nitinol ribbon next to solder, 2000x
	E5417 A-5-b-9	Nitinol ribbon in middle of O.C., 100x
	E5417 A-5-b-10	Nitinol ribbon in middle of O.C., 2000x
	E5417 A-5-b-11	Sandwich weld, 100x
	E5417 A-5-b-12	Sandwich weld, 2000x
	E5417 A-5-b-13	Sandwich weld Solder at $\frac{1}{4}$ turn, 100x
	E5417 A-5-b-14	Solder at $\frac{1}{4}$ turn, transition to inner coil, 2000x

* There was no photo labeled A-5-b-8.

VR 0791

Elisa Aldrich

02-19-02

(JSTB/TW)

2/21/02
Date

PROJECT CORROSION STUDY 90 DAY SEM PHOTOS

Notebook No. 271

Continued From Page 70

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SAMPLE	PHOTO LABEL	DESCRIPTION
A-5-C	E5417 A-5-C-1	Ball tip, 100x
	E5417 A-5-C-2	Ball tip, 2000x
	E5417 A-5-C-3	Inner coils, 4-5 coils closest to solder bond, 100x
	E5417 A-5-C-4	Inner coil, "", 2000x
	E5417 A-5-C-5	Solder band, 100x
	E5417 A-5-C-6	Solder band close to Nitinal ribbon, 2000x
	E5417 A-5-C-7	Nitinal ribbon close to solder bond, 2000x
	E5417 A-5-C-8	Nitinal ribbon in middle of O.C., 100x
	E5417 A-5-C-9	Nitinal ribbon in middle of O.C., 2000x
	E5417 A-5-C-10	Solder tip at 1/3rd of Sandwich weld, 100x
	E5417 A-5-C-11	Sandwich weld, 2000x
	E5417 A-5-C-12	Solder at 1/4 turn, 100x
	E5417 A-5-C-13	Solder at 1/4 turn, transition from solder to inner coil, 2000x



VR 0191

Continued on Page N/A

Read and Understood By

John Alabado

Signed

02-19-02

Date

Signed

02
2/21/02 DT

Date

02
2/21/02

SEM SCIENTIST: JANE WHEELER
 CHARLES EVANS & ASSOCIATES
 APRIL 15, 2002

EA
5/23/02

EACH VIAL CONTAINED 2 SAMPLES
 ONLY 1 SAMPLE FROM EACH VIAL
 WAS IMAGED. GROSS EXAMINATION
 OF THE DEVICES SHOWED NO DIFFERENCE
 BETWEEN THE IMAGED DEVICES AND
 THE DEVICES THAT WERE NOT
 IMAGED. THE DEVICES WERE
 RANDOMLY SELECTED FOR IMAGING.

PURPOSE: GET SEM IMAGES OF 180-DAY CORROSION SAMPLES
 TODAY: IMAGE A-7-A, A-7-B, A-7-C, C-C
 ON LATER DATES, IMAGE C-A, C-B, B-7-A through B-7-E

PROCEDURE: TOOK FULL PHOTOS (14-15 IMAGES) OF SAMPLE
 A-7-A. TOOK IMAGES ONLY OF THE SOLDER FOR THE
 REMAINING SAMPLES.

DATA: IMAGES SAVED ON CD MARKED E5615. HARD COPIES
 WERE ALSO PRINTED.

THE SEM TOOK A LONG TIME (APX 20MIN) TO PULL A VACUUM
 BECAUSE THE SAMPLES WERE NOT COMPLETELY DRY. J. WHEELER
 RECOMMENDED PLACING ANY FUTURE SAMPLES UNDER A HEAT LAMP
 FOR 15-20 MINUTES TO DRY THEM MORE COMPLETELY.

APRIL 22, 2002

PURPOSE: SAME AS ABOVE EXCEPT TODAY IMAGE C-A, C-B, B-7-E,
 B-7-F. DO FULL SET OF IMAGES ON C-A. ALL OTHERS,
 JUST IMAGE SOLDER.

SAMPLE B-7-E OUTER COAT GOT STRETCHED OUT WHEN
 MOUNTING.

DATA: IMAGES SAVED ON CD MARKED E5616 4/22/2002
 HARD COPIES ALSO MADE.

VR 0191

Continued on Page 35

Read and Understood By

Elisa Aldrich ... 5-22-2002

Signed

Date

Signed

Date

(Signature)



5/23/02

PROJECT CORROSION STUDY 180 DAY SEM IMAGES

SEM SCIENTIST: TANE WHEELER
 CHARLES EVANS & ASSOCIATES
 APRIL 22, 2022 11:00-3:00 pm
 30 FA 6.23.2002

PURPOSE: IMAGE SAMPLES B-7-A, B-7-B, B-7-C, B-7-D
 DO FULL SET OF IMAGES FOR SAMPLE B-7-C. FOR THE
 OTHERS, JUST IMAGE SOLDER AND ANY UNUSUAL
 FEATURES, I.E. ANYTHING DIFFERENT FROM B-7-C.

DATA	SAMPLE	FILE NAME	DESCRIPTION	MAGNIFICATION
	B-7-A	a1-E5631	BALL TIP	100x
		a2-E5631	BALL TIP	2000x
		a3-E5631	10/OC SOLDER BOND	100x
		a4-E5631	10/OC SOLDER BOND	2000x
		a5-E5631	STOPPER SOLDER	100x
		a6-E5631	STOPPER SOLDER	2000x

B-7-B

	b1-E5631	BALL TIP	LOW
	b2-E5631	BALL TIP	HIGH
	b3-E5631	10/OC SOLDER BOND	LOW
	b4-E5631	10/OC SOLDER BOND	HIGH
	b5-E5631	STOPPER SOLDER	LOW
	b6-E5631	STOPPER SOLDER	HIGH

B-7-C

	c1-E5631	BALL TIP (GONE)	LOW
	c2-E5631	BALL TIP SOLDER	HIGH
	c3-E5631	IC ^{DIGITAL} PROX. TO SOLDER	LOW
	c4-E5631	IC ^{DIGITAL} PROX. TO SOLDER	HIGH
	c5-E5631	10/OC SOLDER BOND, DIGITAL PROX. END	LOW
	c6-E5631	10/OC SOLDER BOND, PROX. DIGITAL END	LOW

VR 0101

4 dist/prox. redlines/corrections
 20 5-22-2002

Continued on Page 86

Read and Understood By

Elias Alchami

Signed

5-22-2002

Date

John H. Jones

Signed

5/23/02

Date

TA Marseille 2009151 - reçu le 13 janvier 2021 à 16:14 (date et heure de métropole)

<u>TA</u>	<u>SAMPLE</u>	<u>FILENAME</u>	<u>DESCRIPTION</u>	<u>MAGNIFICATION</u>
<u>CONTINUED:</u>				
B-7-C	c-7-e5631	10/OC SOLDER BOND, PROX.	HIGH	
	c8-e5631	NITINOL OC CLOSE TO SOLDER BOND	HIGH	
	c9-e5631	OC, MIDDLE COIL	LOW	
	c10-e5631	OC, MIDDLE COIL	HIGH	
	c11-e5631	STOPPER SOLDER	LOW	(OC ALSO IN IMAGE)
	c12-e5631	STOPPER SOLDER	HIGH	
	c13-e5631	SANDWICH WELD	LOW	
	c14-e5631	SANDWICH WELD	HIGH	
	c15-e5631	SANDWICH WELD	HIGH	
B-7-D	d1-e5631	BALL TIP	LOW	
	d2-e5631	BALL TIP	HIGH	
	d3-e5631	10/OC SOLDER BOND	HIGH	
	d4-e5631	10/OC SOLDER BOND	LOW	
	d5-e5631	STOPPER SOLDER	LOW	
	d6-e5631	STOPPER SOLDER	HIGH	
	d7-e5631	SANDWICH WELD	LOW	
	d8-e5631	SANDWICH WELD	HIGH	
<u>OBSERVATIONS:</u> ON SAMPLE B-7-D, THERE WAS A SPOT NEAR THE SANDWICH WELD THAT APPEARED AS A DIFFERENT COLOR WHEN IMAGED. TANIE WHITLER SAID IT LOOKED LIKE SALT RESIDUE AND NOT CORROSION.				
ALL THE SOLDER SHOWED SIGNS OF CORROSION. THE BALL TIPS CORRODED DRAMATICALLY, THE STOPPER SOLDER SHOWED A LOWER LEVEL OF CORROSION, AND THE 10/OC SOLDER BOND SHOWED THE LEAST CORROSION DAMAGE. NO SAMPLES FELL APART OR HAD 10 OC SEPARATION.				
VR 0191 Read and Understood By _____				
<u>Elia Aldrich</u>		5-22-2002	<u>J. M. H. A.</u>	5/23/02
Signed	Date	Signed	Date	Continued on Page N/A

THE CORROSION SAMPLES ARE FROM CORROSION STUDY:
VR-0191, PV, REV

THE FULL EXPLANATION OF THE CORROSION STUDY DESCRIBES
WHAT EACH SAMPLE IS, BUT A BRIEF DESCRIPTION IS:

IF THE SAMPLE STARTS W/ THE LETTER "A", THE SALINE
IN THE VIAL WAS CHANGED SEVERAL TIMES OVER THE
6 MONTHS. NOT

IF THE SAMPLE STARTS WITH "B", THE SALINE WAS CHANGED

IF THE SAMPLE STARTS WITH "C", IT IS A CONTROL SAMPLE
THAT WAS STORED IN AIR AT ROOM TEMP.

VR 0191		Read and Understood By	Continued on Page
<i>Elsia Almeida</i> Signed	5-23-2002	<i>J. M. Brown</i> Signed	5/23/02

APPENDIX 5

11/28/2001 14:56 4087439798

COALESCENT SURGICAL

PAGE 02/02

From : Scientific Environmental Lab. Inc. C6500 856-5011 Nov. 27, 2001 01:02 PM P02



SCIENTIFIC ENVIRONMENTAL LABORATORIES, INC.

Conceptus, Inc.
1021 Howard Ave,
San Carlos, CA 94070

Mr. Jay Young

Released: 11-27-01
Lab ID : 016836-016865
Rec'd : 11-16-01
Sample: Conceptus
Analyst: DM
Matrix : Liquid

<u>Source</u>	<u>Nickel</u>	<u>Chromium</u>	<u>Tin</u>
A-1 A	0.05	< 0.05	7.9
A-1 B	0.22	< 0.05	41
A-1 C	0.10	< 0.05	12
A-2 A	0.23	< 0.05	38
A-2 B	0.16	< 0.05	59
A-2 C	0.22	< 0.05	50
A-3 A	0.05	< 0.05	2.8
A-3 B	0.35	< 0.05	67
A-3 C	0.12	< 0.05	19
B-1 A	0.17	< 0.05	25
B-1 B	0.24	< 0.05	35
B-1 C	0.23	< 0.05	39
B-1 D	0.22	< 0.05	48
B-1 E	0.18	< 0.05	27
B-1 F	0.31	< 0.05	13
B-2 A	< 0.05	< 0.05	26
B-2 B	< 0.05	< 0.05	28
B-2 C	< 0.05	< 0.05	27
B-2 D	< 0.05	< 0.05	35
B-2 E	< 0.05	< 0.05	29
B-2 F	< 0.05	< 0.05	35
B-3 A	< 0.05	< 0.05	13
B-3 B	< 0.05	< 0.05	2.0
B-3 C	< 0.05	< 0.05	3.6
B-3 D	< 0.05	< 0.05	1.8
B-3 E	< 0.05	< 0.05	3.5
B-3 F	< 0.05	< 0.05	5.9
Control 1 C-D	< 0.05	< 0.05	< 0.05
Control 2 C-E	< 0.05	< 0.05	< 0.05
Control 3 C-F	< 0.05	< 0.05	< 0.05

Nickel - EPA # 249.2 Detection Limit 0.05 mg/L
Chromium - EPA # 218.2 Detection Limit 0.05 mg/L
Tin - EPA # 282.2 Detection Limit 0.05 mg/L

SP:dc

Shui Fong
Shui Fong
Director, Water Laboratory

Banta Certified Water Laboratory for Chemical and Microbiological Examination
924 Industrial Avenue Palo Alto, CA 94303 650 856-5011 FAX 650 856-4281



V R 0191



SCIENTIFIC ENVIRONMENTAL
LABORATORIES, INC.

Conceptus, Inc.
1021 Howard Ave.
San Carlos, CA 94070

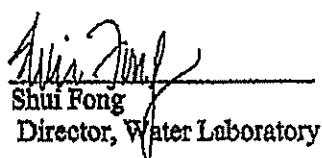
Mr. Jay Young

Released: 12-14-01
Lab ID : 017352-017360
Recv'd : 12-11-01
Sampler: Conceptus
Analyst : DH
Matrix : Liquid

<u>Source</u>	<u>Results (mg/L)</u>		
	<u>Nickel</u>	<u>Chromium</u>	<u>Tin</u>
A-4 A	0.07	<0.05	16.3
A-4 B	<0.05	<0.05	10.8
A-4 C	<0.05	<0.05	4.2
B-4 A	<0.05	<0.05	3.0
B-4 B	<0.05	<0.05	1.2
B-4 C	<0.05	<0.05	1.7
B-4 D	<0.05	<0.05	3.2
B-4 E	<0.05	<0.05	4.8
B-4 F	<0.05	<0.05	2.7

Nickel - EPA # 249.2 Detection Limit 0.05 mg/L
Chromium - EPA # 218.2 Detection Limit 0.05 mg/L
Tin - EPA # 282.2 Detection Limit 0.05 mg/L

SF:dc


Shui Fong
Director, Water Laboratory

VR 0191

State Certified Water Laboratory for Chemical and Biological Examination
924 Industrial Avenue Palo Alto, CA 94303 650 856-6011 FAX 650 856-4281





**SCIENTIFIC ENVIRONMENTAL
LABORATORIES, INC.**

Conceptus, Inc.
1021 Howard Ave.
San Carlos, CA 94070

Mr. Jay Yang

Released: 1-24-02
Lab ID : 020332-020340
Recv'd : 1-18-02
Sampler: Conceptus
Analyst : DH
Matrix : Liquid

<u>Source</u>	<u>Results (mg/L)</u>		
	<u>Nickel</u>	<u>Chromium</u>	<u>Tin</u>
A-5 A	<0.05	<0.05	2.2
A-5 B	<0.05	<0.05	4.5
A-5 C	<0.05	<0.05	4.4
B-5 A	<0.05	<0.05	5.1
B-5 B	<0.05	<0.05	1.4
B-5 C	<0.05	<0.05	0.5
B-5 D	<0.05	<0.05	0.7
B-5 E	<0.05	<0.05	0.9
B-5 F	<0.05	<0.05	2.0

Nickel - EPA # 249.2 Detection Limit 0.05 mg/L
Chromium - EPA # 218.2 Detection Limit 0.05 mg/L
Tin - EPA # 282.2 Detection Limit 0.05 mg/L

SF:do


Shui Fong
Director, Water Laboratory

VR 0191

State Certified Water Laboratory for Chemical and Biological Examination
924 Industrial Avenue Palo Alto, CA 94303 (650) 856-6011 FAX (650) 856-4281





SCIENTIFIC ENVIRONMENTAL LABORATORIES, INC.

Conceptus, Inc.
1021 Howard Ave.
San Carlos, CA 94070

Ms. Elisa Aldridge

Released: 2-28-02
Lab ID : 020941-020949
Recv'd : 2-22-02
Sampler: Conceptus
Analyst : DH
Matrix : Liquid

<u>Source</u>	<u>Results (mg/L)</u>		
	<u>Nickel</u>	<u>Chromium</u>	<u>Tin</u>
A-6 A	0.10	< 0.05	17.6
A-6 B	0.05	< 0.05	7.0
A-6 C	0.05	< 0.05	6.3
B-6 A	< 0.05	< 0.05	0.69
B-6 B	< 0.05	< 0.05	0.33
B-6 C	< 0.05	< 0.05	< 0.05
B-6 D	< 0.05	< 0.05	0.15
B-6 E	< 0.05	< 0.05	0.61
B-6 F	< 0.05	< 0.05	17.6

Nickel - EPA # 249.2 Detection Limit 0.05 mg/L

Chromium - EPA # 218.2 Detection Limit 0.05 mg/L

Tin - EPA # 282.2 Detection Limit 0.05 mg/L

SF:dc


Shui Fong
Director, Water Laboratory

V R 0191



State Certified Water Laboratory for Chemical and Biological Examination
924 Industrial Avenue Palo Alto, CA 94303 650 856-8011 FAX 650 856-4281



**SCIENTIFIC ENVIRONMENTAL
LABORATORIES, INC.**

Conceptus, Inc.
1021 Howard Ave.
San Carlos, CA 94070

Ms. Elisa Aldridge

Released: 4-23-02
Lab ID : 021777-021785
Reov'd : 4-11-02
Sampler : Conceptus
Analyst : DH
Matrix : Liquid
Revised

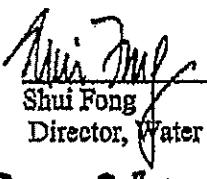
<u>Source</u>	<u>Results (mg/L)</u>		
	<u>Nickel</u>	<u>Chromium</u>	<u>Tin</u>
A-7 A	0.04	< 0.05	5.2
A-7 B	0.14	< 0.05	5.9
A-7 C	0.40	< 0.05	11.6
B-7 A	0.03	< 0.05	0.09
B-7 B	0.05	< 0.05	0.05
B-7 C	0.02	< 0.05	0.07
B-7 D	0.05	< 0.05	0.05
B-7 E	0.05	< 0.05	0.14
B-7 F	0.03	< 0.05	0.60

Nickel - EPA # 249.2 Detection Limit 0.02 mg/L

Chromium - EPA # 218.2 Detection Limit 0.05 mg/L

Tin - EPA # 282.2 Detection Limit 0.05 mg/L

SF:dc


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VR 0191

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Attachment

Corrosion Susceptibility Test Protocol on Conceptus Essure™ pbc device	V P0191.pv rev1	
	10/11/01	

Corrosion Susceptibility Test Protocol Conceptus Essure™ pbc device

Approvals:

Document Originator

Jay Yang, PhD
Consultant
Global Nitinol Technologies

Date: 10.12.01

Research and Development

Date: 10.12.01

Regulatory Affairs

Date: 10/12/2001



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Objective

To evaluate the corrosion susceptibility of the implantable Essure™ pbc device.

Background

Corrosion and its effects on structural integrity and biocompatibility are important part of design validation on efficacy and safety of implantable medical devices, especially when the device is made of dissimilar metals. Since an implanted Essure™ pbc device is made of several types of materials (Tin solder, nitinol, 316L stainless steel, platinum), it is recommended to conduct a corrosion test to assess potential risks associated with galvanic corrosion.

Scope

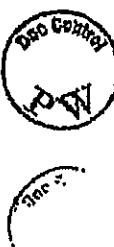
1. To determine maximum leaching rate of toxic metal ions (Ni, Sn, Cr, specifically) in a simulated body environment.
2. To determine minimum time needed for structural disintegration /detachment in a simulated body environment.

Highlights of Methods

1. Samples: n=60 actual devices (implantable portion only) with current validated processes.
2. Simulated environment: immersion in a 0.9% NaCl physiological saline solution at 37 +/- 1°C for up to 6 months. The pH value of the saline will be buffered between 6.0 and 7.0 at the beginning of the test. Corrosion data will be collected both in a cumulative sampling method (individual sample groups will be used for selected time points and solutions will not be changed during the sampling periods) and in a disruptive sampling method (same sample groups will be used for all selected time points and only solutions will be changed at the selected time points).
3. Analytic methods: (a) Using atomic absorption spectroscopy to measure leaching rate of metal ions (Ni, Sn, Cr) periodically for up to 6 months; and (b) use scanning electron microscopy (SEM) to record surface condition and components integrity periodically for up to 6 months.

Reference Documents

1. ASTM F746-87: "Standard Test Method for Pitting or Crevice Corrosion of Metallic Surgical Implant Materials"
2. "Medical and Biological Effects of Environmental Pollutants - Nickel", published by the National Academy of Sciences, Washington, D.C., 1975.
3. Measurement of trace element Nickel by atomic absorption spectroscopy, test method EPA 249.2



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4. Measurement of trace element Chromium by atomic absorption spectroscopy, test method EPA 218.2
5. Measurement of trace element Tin by atomic absorption spectroscopy, test method EPA 282.2

Materials and Equipment

Implantable devices, p/n E0648-02 , Qty 60
 Water bath with temperature control at +/-1°C.

Distilled water

Optional Polypropylene balls, hollow

Physiological saline solution, 0.9% NaCl

Buffer solutions pH 6.0, pH 7.0, pH 8.0, pH 5.0

Glass vials, 4 ml, with Teflon sealed screw top

Vial rack or glass containers with screw top

Pipette and suction bulb

Electronic temperature gage (calibrated, see report for calibration information)

Procedure

1. Fill the water bath with distilled water.
2. Add polypropylene balls to top surface of water in the bath so that the entire top surface is covered (this will reduce evaporation of water from the bath).
3. Set water bath temperatures to 37° Celsius.
4. Allow baths to heat for 2 hours to allow temperatures to stabilize.
5. Confirm temperature of bath; recalibrate temperature control of bath if necessary using procedure detailed in instruction manual of bath. Check and note temperature of bath at minimum once a week. Check water level at least once a week, to make sure vials are between 50% and 80% immersed in the water of the bath.
6. Prepare thirty (30) glass vials and divide them into three groups (A, B, C) by marking each as follows with an indelible ink marker:

A-1-a:	7 days, cumulative
A-1-b:	7 days, cumulative
A-1-c:	7 days, cumulative
A-2-a:	14 days, cumulative
A-2-b:	14 days, cumulative
A-2-c:	14 days, cumulative
A-3-a:	30 days, cumulative
A-3-b:	30 days, cumulative
A-3-c:	30 days, cumulative
A-4-a:	60 days, cumulative
A-4-b:	60 days, cumulative
A-4-c:	60 days, cumulative
A-5-a:	90 days, cumulative
A-5-b:	90 days, cumulative

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A-5-c:	90 days, cumulative
A-6-a:	120 days, cumulative
A-6-b:	120 days, cumulative
A-6-c:	120 days, cumulative
A-7-a:	180 days, cumulative
A-7-b:	180 days, cumulative
A-7-c:	180 days, cumulative
B-x-a:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
B-x-b:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
B-x-c:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
B-x-d:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
B-x-e:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
B-x-f:	x=1,2,3,4,5,6,7, representing 7, 14, 30, 60, 90, 120, 180 days
C-a:	Control / reference, device only
C-b:	Control / reference, device only
C-c:	Control / reference, device only
C-d:	Control / reference, solution only
C-e:	Control / reference, solution only
C-f:	Control / reference, solution only

7. Prepare Sixty (60) implantable devices using current engineering procedures, including the heat treatment, cleaning, and inspection criteria. Sterilization is not necessary. Record complete history of the devices.
8. Take scanning electron micrographs (SEM's) of six (6) devices as control or reference. Micrographs should be taken at both lower magnification and high magnification at representative locations: solder joint, weld joint, inner coil, and outer coil.
9. Place two (2) of the devices into each of the all vials, except for c-d, c-e, c-f.
10. Add minimum amount of the pH 6.0 buffer solution to set the saline solution at pH between 6.0 and 7.0.
11. Using the pipette and suction bulb, add sufficient physiological saline solution to the first vial to fully cover all devices. Note exact amount added. Close screw cap tightly.
12. Add this same volume of saline, measured with the pipette, to all other glass vials except for the C-a, C-b and C-c vials. Close all vial lids tightly. Send solution control samples C-d, C-e, C-f to a contract lab for trace elements (Ni, Sn, Cr) analysis.
13. Place the vials in water bath set to 37°C. The vials may need to be weighted down or fixed onto a rack to keep it submerged. Replace cover on water bath and note date and time.
14. Keep the three vials labeled "control" at room temperature in a cool place. Keep the lids tightly closed. No temperature control is required.
15. At 7 days after the start of the test, remove the vials labeled "7 days" in sample group A from the water bath. Remove all devices from each vial; ultrasonically clean the

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devices and place them in new, labeled vials. Send the aged solution of each vial to a contract lab for trace elements (Ni, Sn, Cr) analysis and the devices of each vial for SEM analysis. SEM Micrographs should be taken both at lower magnification and higher magnification at each representative locations: solder joint, weld joint, inner coil, and outer coil, with focuses on specific corrosion signs such as pits, cracking, massive erosion, and detachment of joints.

16. Similarly, at 7 days after the start of the test, remove the vials labeled "7 days" in sample group B from the water bath. Remove all devices from each vial and place them in new, labeled vials. Fill the new vials with the same amount (4 ml) of fresh buffered saline solution and close the lids tightly. Place the new vials back into the water bath. Send the aged solutions to a contract lab for trace elements (Ni, Sn, Cr) analysis.
17. Repeat Steps 15 and 16 at 14 days.
18. Repeat Steps 15 and 16 at 30 days.
19. Repeat Steps 15 and 16 at 60 days. In addition, perform SEM analysis on devices from sample group B. Ultrasonically clean the devices before and after SEM handling.
20. Repeat Steps 15 and 16 at 90 days.
21. Repeat Steps 15 and 16 at 120 days.
22. Repeat Steps 15 and 16 at 180 days. In addition, perform SEM analysis on devices from sample group B. Ultrasonically clean the devices before and after SEM handling
23. Analyze data and write an intermediate report at about 2-months of data points and a final report of findings when the test ends.

Acceptance Criteria

1. The measured maximum corrosion rate must be such that the resulting amount of equivalent nickel release is less than the normal human intake of nickel from the natural environment, which is 0.3 to 0.6 mg per day, as published in reference 2.
2. There is no complete detachment of the components caused by corrosion within first 3 months of the tested time periods.

Results and Conclusions

All build documentation for the samples, calibration information and test data will be included in a final report.

The results and conclusions of all tests shall be documented in a final report.

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Attachment 2

Addendum to Corrosion Susceptibility Test Protocol on Conceptus Essure Device	VP-0191.PV.Rev1 A01	Page 1/2
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**Addendum to
Corrosion Susceptibility Test Protocol
On Conceptus Essure Device**

VP-0191.PV.Rev1 A01

Approvals:

Document Originator	Elisa Aldridge	<u>Elisa Aldridge</u>	Date: 2-22-02
Research and Development	Ashish Khera	<u>Ashish Khera</u>	Date: 2-22-02
Regulatory Affairs	Susan Aloyan	<u>Susan Aloyan</u>	Date: 3-4-2002

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Summary

This addendum describes and approves a deviation from the original protocol, VP-0191.PV.Rev1. An uncalibrated water bath may be used for the validation. This is an acceptable deviation, because the water temperature is being monitored with a calibrated electronic temperature gage.

Original Protocol Requirements

1. Required equipment includes:
 - 1.1 Water bath with temperature control at $\pm 1^\circ\text{C}$
 - 1.2 Electronic temperature gage
 - 1.3 Unless specified otherwise in a validation protocol, only calibrated equipment may be used in a validation.
2. Step 5 of the procedure includes:
 - 2.1 Confirm the temperature of the bath is $37 \pm 1^\circ\text{C}$ before setting the samples in the water bath.
 - 2.2 Check and note temperature of bath at minimum once a week.

Deviation

VP-0191.PV.Rev1 may be performed using an uncalibrated water bath as long as a calibrated temperature gage is used to monitor the temperature of the water bath. The original protocol did not clearly state that an uncalibrated water bath is acceptable.

Justification

1. The temperature is being monitored at least once a week using a calibrated electronic temperature gage.
2. During the 4 months that the test has been running, the water bath temperature has varied from 36.1°C to 37.8°C . This is within the range required by the protocol. The data is located in lab notebook # 239, pages 64 and 65.

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